

Quinoxaline based derivatives from the reaction of 2-azabicyclo-[3.3.0]octa-2,7-dien-4,6-dione-N-oxide with *o*-phenylenediamines

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**Dedicated to Prof. Alexander F. Pozharskii on the occasion of his 70th Birthday
in recognition of his outstanding contribution to the chemistry of heterocyclic compounds
and physical organic chemistry**

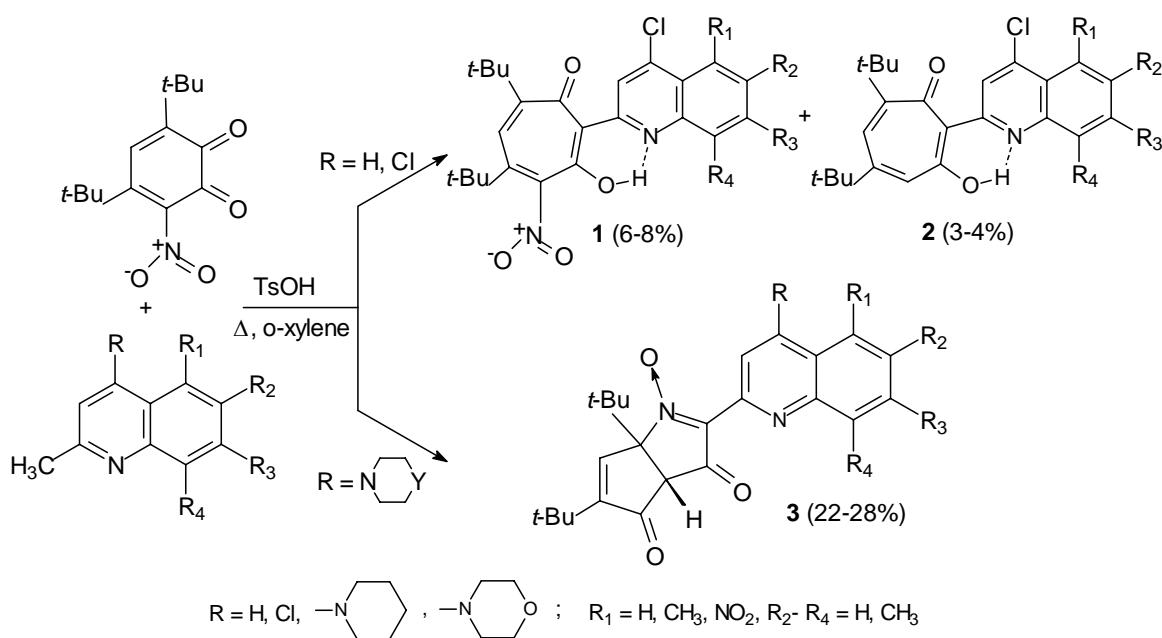
Abstract

Novel hetaryl substituted quinoxalines have been prepared by the acid-catalyzed condensation of 2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-*N*-oxide with derivatives of *o*-phenylenediamine. The mechanism of the reaction has been considered and the molecular structure of one of the prepared quinoxalines determined by X-ray crystallography.

Keywords: 3,5-Di-(*tert*-butyl)-1,2-benzo-quinone, *o*-phenylenediamines, quinoxalines, crystal structure

Introduction

We have recently reported that the reaction of 3-nitro-4,6-di(*tert*-butyl)-1,2-benzoquinone with 2-methylquinolines occurring upon heating under reflux of their *o*-xylene solution for 1 h leads to the expansion of the six-membered *o*-quinone ring resulting in the formation of 2-quinolinyl derivatives of 4-nitro-β-tropolones **1** and 2-quinolinyl-β-tropolones **2** as the minor products (Scheme 1).¹

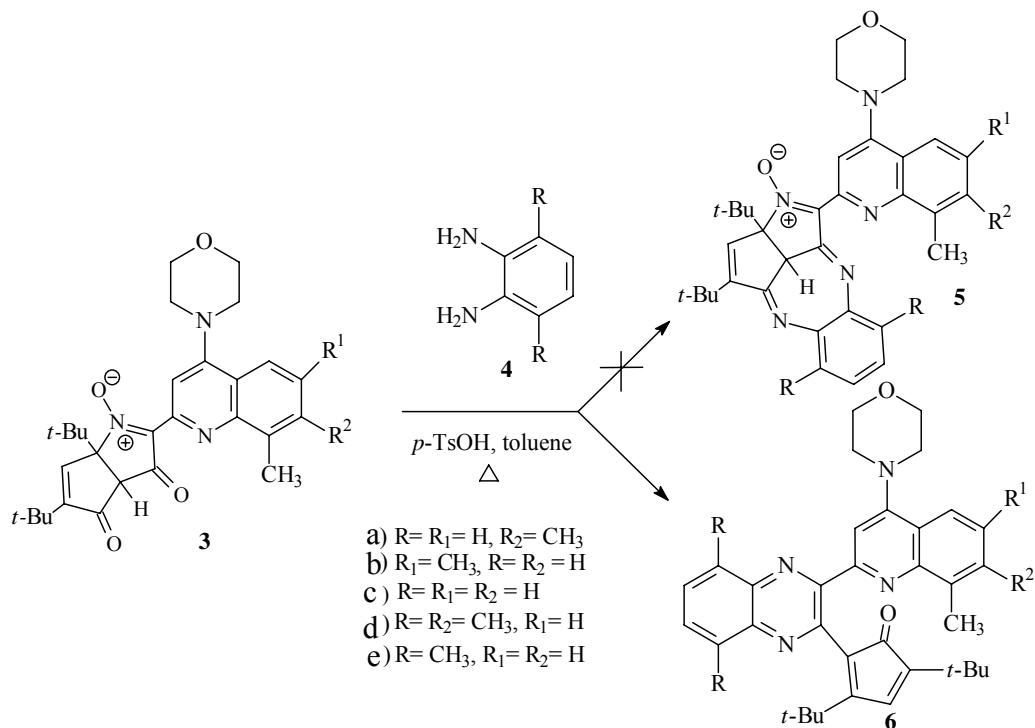
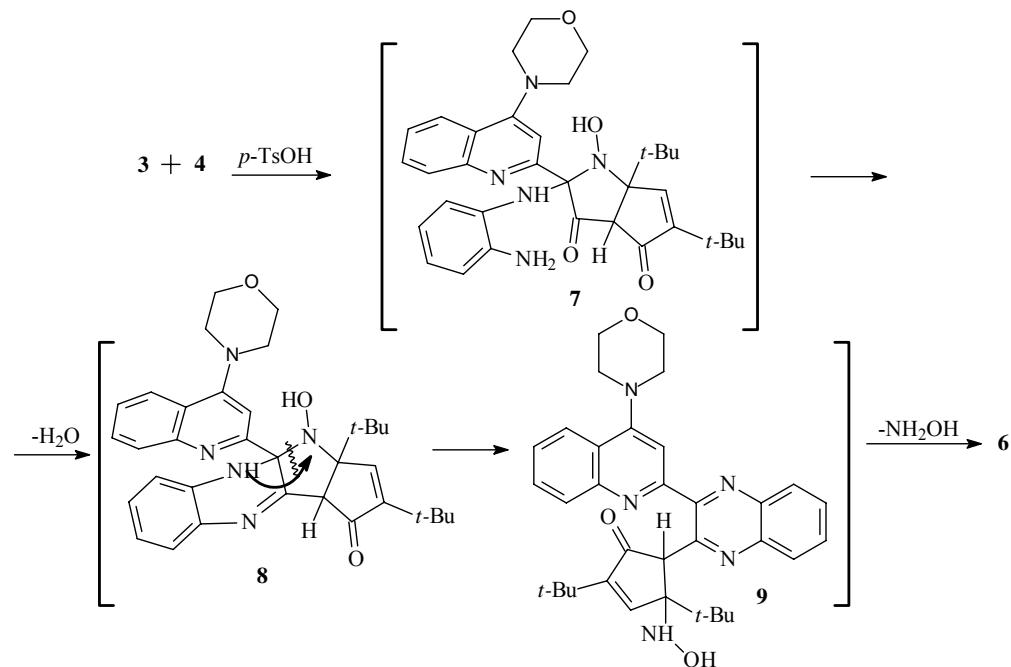
**Scheme 1**

Under the same conditions 2-methylquinolines containing a strong electron-releasing group ($\text{N}(\text{CH}_2)_4\text{O}$, $\text{N}(\text{CH}_2)_5$) in the position 4 react differently affording readily isolated crystal products, the structure of which was identified with the use of X-ray crystallography as 3-(2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-*N*-oxides **3**.¹⁻³ Since the molecular framework of compounds **3** contains a 1,3-diketone moiety one may expect that these will be prone to some of the reactions characteristic of 1,3-diketones⁴. In the hope of preparing new derivatives of 1,5-benzodiazepines we have studied the condensation of compounds **3** with *o*-phenylenediamines.

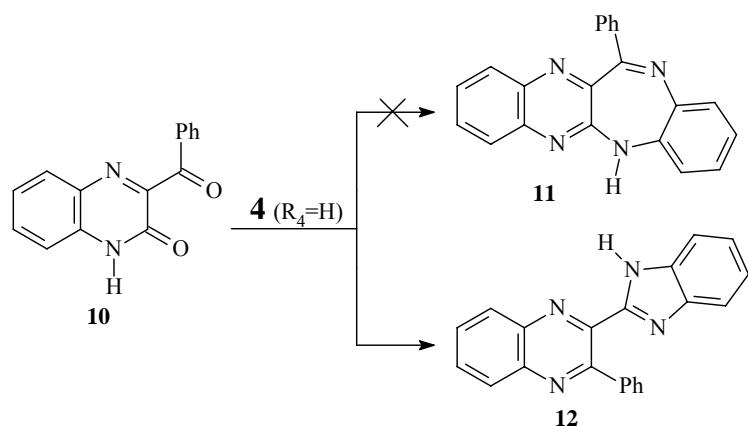
Results and Discussion

Contrary to expectations the acid-catalyzed reaction of 3-(2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-*N*-oxides **3** with *o*-phenylenediamines gives rise not to 1,5-benzodiazepines **5**, but to previously unknown hetaryl derivatives of quinoxaline **6**, obtained in 57-74 % yields (Scheme 2).

The conjectural mechanism of the reaction is shown in Scheme 3. At the initial stage a molecule of *o*-phenylenediamine adds to the electrophilic C(3) center of **3** that is activated by the adjacent *N*-oxide group.^{5,6} The subsequent intramolecular cyclization of **7** proceeding with elimination of a molecule of water affords the dihydroquinoxaline **8**, which undergoes 1,3-sigmatropic shift of a hydrogen **8**→**9**. The elimination of a molecule of hydroxylamine then results in **6**.

**Scheme 2****Scheme 3**

A previous attempt⁷ to synthesize derivatives of 1,5-benzodiazepine by coupling *o*-phenylenediamine with 3-benzoyl-1,2-dihydro-2-oxoquinoxaline **10** containing an 1,3-dicarbonyl fragment has also failed. The prolonged heating of an acetic acid solution of the components gave rise not to the expected quinoxalinobenzodiazepine **11**, but to its isomer, 2-benzimidazolyl-3-phenylquinoxaline **12** (Scheme 4). The mechanism of this reaction is assumed to involve a complicated transformation of the quinoxaline ring of **10** into the 2-benzimidazolyl substituent, whereas the 3-phenylquinoxalinyll fragment of **12** is formed from *o*-phenylenediamine.



Scheme 4

The majority of currently known methods for the synthesis of quinoxalines are based on the reactions of *o*-phenylenediamines with 1,2-dicarbonyl compounds or their oximes.^{7,8} Therefore, the method for the preparation of compounds **6** using 1,3-dicarbonyl-containing compounds **3** can be regarded as a new approach to the derivatives of quinoxaline. The molecular structure of the compound, **6a** ($R^4 = R^2 = H$, $R^3 = CH_3$) has been proved by X-ray analysis (Figure 1). The values of important bond lengths and valence angles are given in Table 1.

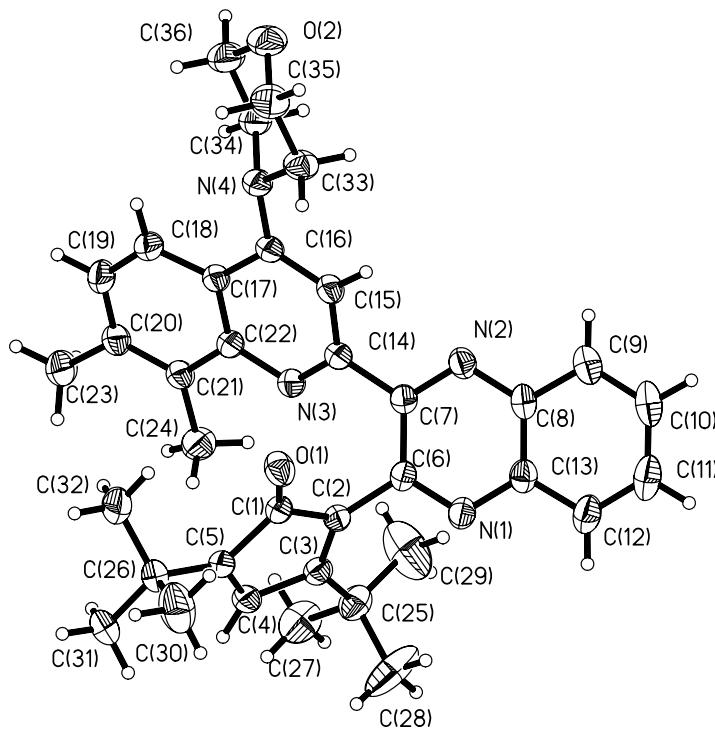


Figure 1. Molecular structure of 3,5-di(*tert*-butyl)-2-[3-(7,8-dimethyl-4-morpholino-2-quinolyl)-2-quinoxalinyl]-2,4-cyclopentadien-1-one (**6a**).

Table 1. Bond lengths [Å] and angles [deg] for (**6a**)

N(1)-C(6)	1.323(3)	N(1)-C(13)	1.366(4)
N(2)-C(7)	1.317(3)	N(2)-C(8)	1.377(3)
N(3)-C(14)	1.322(3)	N(3)-C(22)	1.381(3)
N(4)-C(16)	1.409(3)	N(4)-C(33)	1.452(3)
N(4)-C(34)	1.470(4)	O(1)-C(1)	1.217(3)
O(2)-C(36)	1.415(4)	O(2)-C(35)	1.419(4)
C(1)-C(2)	1.498(4)	C(1)-C(5)	1.510(4)
C(2)-C(3)	1.341(4)	C(2)-C(6)	1.475(4)
C(3)-C(4)	1.502(4)	C(3)-C(25)	1.510(4)
C(4)-C(5)	1.326(4)	C(5)-C(26)	1.508(4)
C(6)-C(7)	1.434(4)	C(7)-C(14)	1.494(4)
C(8)-C(13)	1.392(4)	C(8)-C(9)	1.411(4)
C(9)-C(10)	1.379(4)	C(10)-C(11)	1.384(5)
C(11)-C(12)	1.354(5)	C(12)-C(13)	1.416(4)
C(14)-C(15)	1.404(4)	C(15)-C(16)	1.376(3)

Table 1. Continued

C(17)-C(18)	1.415(4)	C(18)-C(19)	1.360(4)
C(19)-C(20)	1.407(4)	C(20)-C(21)	1.377(4)
C(20)-C(23)	1.510(4)	C(21)-C(22)	1.429(4)
C(21)-C(24)	1.500(4)	C(25)-C(28)	1.519(5)
C(25)-C(27)	1.521(5)	C(25)-C(29)	1.531(4)
C(26)-C(32)	1.514(4)	C(26)-C(31)	1.517(4)
C(26)-C(30)	1.541(5)	C(33)-C(35)	1.496(4)
C(34)-C(36)	1.506(4)	C(6)-N(1)-C(13)	117.7(3)
C(7)-N(2)-C(8)	117.3(2)	C(14)-N(3)-C(22)	117.7(2)
C(16)-N(4)-C(33)	117.3(2)	C(16)-N(4)-C(34)	115.3(2)
C(33)-N(4)-C(34)	109.3(2)	C(36)-O(2)-C(35)	109.6(3)
O(1)-C(1)-C(2)	126.1(3)	O(1)-C(1)-C(5)	126.7(3)
C(2)-C(1)-C(5)	107.2(2)	C(3)-C(2)-C(6)	133.0(3)
C(3)-C(2)-C(1)	107.7(2)	C(6)-C(2)-C(1)	118.9(2)
C(2)-C(3)-C(4)	107.5(3)	C(2)-C(3)-C(25)	131.2(3)
C(4)-C(3)-C(25)	121.3(3)	C(5)-C(4)-C(3)	112.9(3)
C(4)-C(5)-C(26)	130.5(3)	C(4)-C(5)-C(1)	104.7(2)
C(26)-C(5)-C(1)	124.6(3)	N(1)-C(6)-C(7)	121.1(3)
N(1)-C(6)-C(2)	115.4(3)	C(7)-C(6)-C(2)	123.2(2)
N(2)-C(7)-C(6)	121.7(2)	N(2)-C(7)-C(14)	116.2(3)
C(6)-C(7)-C(14)	122.1(2)	N(2)-C(8)-C(13)	121.0(3)
N(2)-C(8)-C(9)	118.9(3)	C(13)-C(8)-C(9)	120.0(3)
C(10)-C(9)-C(8)	118.8(3)	C(9)-C(10)-C(11)	120.6(3)
C(12)-C(11)-C(10)	121.7(3)	C(11)-C(12)-C(13)	119.2(3)
N(1)-C(13)-C(8)	121.1(2)	N(1)-C(13)-C(12)	119.1(3)
C(8)-C(13)-C(12)	119.7(3)	N(3)-C(14)-C(15)	123.7(2)
N(3)-C(14)-C(7)	117.0(2)	C(15)-C(14)-C(7)	119.3(3)
C(16)-C(15)-C(14)	120.2(3)	C(15)-C(16)-N(4)	122.9(2)
C(15)-C(16)-C(17)	117.7(2)	N(4)-C(16)-C(17)	119.3(2)
C(22)-C(17)-C(18)	117.9(2)	C(22)-C(17)-C(16)	118.4(2)
C(18)-C(17)-C(16)	123.7(3)	C(19)-C(18)-C(17)	119.8(3)
C(18)-C(19)-C(20)	123.1(3)	C(21)-C(20)-C(19)	118.7(3)
C(21)-C(20)-C(23)	122.3(3)	C(19)-C(20)-C(23)	119.0(3)
C(20)-C(21)-C(22)	119.3(3)	C(20)-C(21)-C(24)	121.9(3)
C(22)-C(21)-C(24)	118.7(2)	N(3)-C(22)-C(17)	122.2(2)
N(3)-C(22)-C(21)	116.8(2)	C(17)-C(22)-C(21)	121.0(2)
C(3)-C(25)-C(28)	107.8(3)	C(3)-C(25)-C(27)	111.6(3)
C(28)-C(25)-C(27)	112.0(4)	C(3)-C(25)-C(29)	110.3(3)
C(28)-C(25)-C(29)	107.8(3)	C(27)-C(25)-C(29)	107.4(3)

Table 1. Continued

C(5)-C(26)-C(32)	109.8(3)	C(5)-C(26)-C(31)	110.9(3)
C(32)-C(26)-C(31)	109.2(3)	C(5)-C(26)-C(30)	108.0(2)
C(32)-C(26)-C(30)	110.1(3)	C(31)-C(26)-C(30)	108.8(3)
N(4)-C(33)-C(35)	109.3(3)	N(4)-C(34)-C(36)	109.0(3)
O(2)-C(35)-C(33)	112.0(3)	O(2)-C(36)-C(34)	112.3(3)
O(1)-C(1)-C(2)-C(3)	176.2(3)	C(5)-C(1)-C(2)-C(3)	-0.2(3)
O(1)-C(1)-C(2)-C(6)	1.8(4)	C(5)-C(1)-C(2)-C(6)	-174.7(2)
C(6)-C(2)-C(3)-C(4)	172.6(3)	C(1)-C(2)-C(3)-C(4)	-0.7(3)
C(6)-C(2)-C(3)-C(25)	6.7(5)	C(1)-C(2)-C(3)-C(25)	179.9(3)
C(2)-C(3)-C(4)-C(5)	1.6(3)	C(25)-C(3)-C(4)-C(5)	-179.0(3)
C(3)-C(4)-C(5)-C(26)	-177.6(3)	C(3)-C(4)-C(5)-C(1)	-1.7(3)
O(1)-C(1)-C(5)-C(4)	-175.2(3)	C(2)-C(1)-C(5)-C(4)	1.2(3)
O(1)-C(1)-C(5)-C(26)	1.0(4)	C(2)-C(1)-C(5)-C(26)	177.4(2)
C(13)-N(1)-C(6)-C(7)	-0.6(4)	C(13)-N(1)-C(6)-C(2)	-174.3(2)
C(3)-C(2)-C(6)-N(1)	-58.0(4)	C(1)-C(2)-C(6)-N(1)	114.8(3)
C(3)-C(2)-C(6)-C(7)	128.4(3)	C(1)-C(2)-C(6)-C(7)	-58.8(3)
C(8)-N(2)-C(7)-C(6)	0.5(4)	C(8)-N(2)-C(7)-C(14)	-179.4(2)
N(1)-C(6)-C(7)-N(2)	-0.5(4)	C(2)-C(6)-C(7)-N(2)	172.7(3)
N(1)-C(6)-C(7)-C(14)	179.4(3)	C(2)-C(6)-C(7)-C(14)	-7.4(4)
C(7)-N(2)-C(8)-C(13)	0.5(4)	C(7)-N(2)-C(8)-C(9)	-179.2(3)
N(2)-C(8)-C(9)-C(10)	179.7(3)	C(13)-C(8)-C(9)-C(10)	-0.1(4)
C(8)-C(9)-C(10)-C(11)	-0.7(5)	C(9)-C(10)-C(11)-C(12)	0.7(5)
C(10)-C(11)-C(12)-C(13)	0.2(5)	C(6)-N(1)-C(13)-C(8)	1.6(4)
C(6)-N(1)-C(13)-C(12)	178.8(3)	N(2)-C(8)-C(13)-N(1)	-1.6(4)
C(9)-C(8)-C(13)-N(1)	178.2(3)	N(2)-C(8)-C(13)-C(12)	-178.8(3)
C(9)-C(8)-C(13)-C(12)	1.0(4)	C(11)-C(12)-C(13)-N(1)	-178.2(3)
C(11)-C(12)-C(13)-C(8)	-1.0(5)	C(22)-N(3)-C(14)-C(15)	-2.4(4)
C(22)-N(3)-C(14)-C(7)	179.2(2)	N(2)-C(7)-C(14)-N(3)	148.2(3)
C(6)-C(7)-C(14)-N(3)	-31.7(4)	N(2)-C(7)-C(14)-C(15)	-30.2(4)
C(6)-C(7)-C(14)-C(15)	149.9(3)	N(3)-C(14)-C(15)-C(16)	2.7(4)
C(7)-C(14)-C(15)-C(16)	-179.0(2)	C(14)-C(15)-C(16)-N(4)	178.0(3)
C(14)-C(15)-C(16)-C(17)	0.6(4)	C(33)-N(4)-C(16)-C(15)	-15.8(4)
C(34)-N(4)-C(16)-C(15)	115.1(3)	C(33)-N(4)-C(16)-C(17)	161.6(3)
C(34)-N(4)-C(16)-C(17)	-67.5(3)	C(15)-C(16)-C(17)-C(22)	-3.8(4)
N(4)-C(16)-C(17)-C(22)	178.7(2)	C(15)-C(16)-C(17)-C(18)	173.9(3)
N(4)-C(16)-C(17)-C(18)	-3.6(4)	C(22)-C(17)-C(18)-C(19)	-3.8(4)
C(16)-C(17)-C(18)-C(19)	178.4(3)	C(17)-C(18)-C(19)-C(20)	-0.1(5)
C(18)-C(19)-C(20)-C(21)	3.3(5)	C(18)-C(19)-C(20)-C(23)	-178.1(3)
C(19)-C(20)-C(21)-C(22)	-2.4(4)	C(23)-C(20)-C(21)-C(22)	179.0(3)

Table 1. Continued

C(19)-C(20)-C(21)-C(24)	177.7(3)	C(23)-C(20)-C(21)-C(24)	-0.8(5)
C(14)-N(3)-C(22)-C(17)	-1.1(4)	C(14)-N(3)-C(22)-C(21)	-179.5(2)
C(18)-C(17)-C(22)-N(3)	-173.7(3)	C(16)-C(17)-C(22)-N(3)	4.2(4)
C(18)-C(17)-C(22)-C(21)	4.6(4)	C(16)-C(17)-C(22)-C(21)	-177.5(3)
C(20)-C(21)-C(22)-N(3)	176.9(3)	C(24)-C(21)-C(22)-N(3)	-3.3(4)
C(20)-C(21)-C(22)-C(17)	-1.5(4)	C(24)-C(21)-C(22)-C(17)	178.4(3)
C(2)-C(3)-C(25)-C(28)	99.8(4)	C(4)-C(3)-C(25)-C(28)	-79.5(4)
C(2)-C(3)-C(25)-C(27)	-23.6(5)	C(4)-C(3)-C(25)-C(27)	157.1(3)
C(2)-C(3)-C(25)-C(29)	-142.8(3)	C(4)-C(3)-C(25)-C(29)	37.9(4)
C(4)-C(5)-C(26)-C(32)	-132.8(3)	C(1)-C(5)-C(26)-C(32)	52.0(4)
C(4)-C(5)-C(26)-C(31)	-12.1(4)	C(1)-C(5)-C(26)-C(31)	172.8(3)
C(4)-C(5)-C(26)-C(30)	107.1(4)	C(1)-C(5)-C(26)-C(30)	-68.1(4)
C(16)-N(4)-C(33)-C(35)	-167.9(3)	C(34)-N(4)-C(33)-C(35)	58.5(3)
C(16)-N(4)-C(34)-C(36)	167.8(2)	C(33)-N(4)-C(34)-C(36)	-57.5(3)
C(36)-O(2)-C(35)-C(33)	57.9(4)	N(4)-C(33)-C(35)-O(2)	-59.3(4)
C(35)-O(2)-C(36)-C(34)	-57.3(4)	N(4)-C(34)-C(36)-O(2)	57.7(4)

The quinoline ring of **6a** is virtually planar. Deviations of C(14)-C(22) and N(3) atoms from the mean-square plane are in the limits of 0.063(2) to -0.053(2) Å, whereas those of C(23) and C(24) atoms are 0.069(4) and 0.075(4) Å, respectively. The largest deviations from the mean-square plane found for atoms C(7) and N(4) are equal to 0.137(4) and 0.094(4) Å, respectively. A morpholine ring in **6a** having the chair conformation is turned around the N(4)-C(16) bond in such a way that the dihedral angles C(33)-N(4)-C(16)-C(17) and C(34)-N(4)-C(16)-C(17) are equal to 119.3(3)° and -67.5(3)°, respectively. The sum of the valence angles at the tetrahedral N(4) centre is equal to 342.8°. The N(1), N(2), C(6)-C(12) atoms of the quinoxaline moiety of **6a** deviate from the common mean-square plane in the range of -0.010(2) - 0.022(3) Å. This plane contains also the C(4) atom. The torsion angle N(2)-C(7)-C914)-N(3) is equal to 148.2°. Deviations of atoms O(1), C(25), C(25) and C(6) from the mean-square plane of the cyclopentadienyl ring are -0.075(4), 0.014(5), -0.047(5) and -0.132(4) Å, respectively. The dihedral angles C(1)-C(5)-C(26)-C(31), C(4)-C(3)-C(25)-C(27) and C(1)-C(2)-C(6)-C(7) are 172.8(3)°, 157.1(3)° and -58.8(4)°, respectively. The crystal structure of **6a** is characterized by a shortened contact (2.83 Å) between N(1) atom of one molecule and a hydrogen atom of an adjacent molecule.

Experimental Section

General Procedures. The elemental analyses were carried out by the laboratory of microanalysis of the Institute of Physical and Organic Chemistry. The ¹H NMR spectra were recorded on a Varian Unity-300 spectrometer. The mass spectra were run on a Finnigan MAT

INCOS-50 instrument. The IR spectra were measured on a Specord IR-75 spectrometer from samples dispersed in nujol mulls. The chromatography was performed using standard aluminium oxide columns. Melting temperatures (uncorrected) were measured in glass capillaries with the use of a PTP instrument.

X-Ray analysis

The unit cell parameters and reflection intensities (a three-dimensional set) were measured on a Bruker P-4 autodiffractometer (λ MoK α irradiation, graphite monochromator). Monoclinic crystals; molecular formula C₃₆H₄₀O₂N₄, M = 560.72; *a* 16.721(2), *b* 12.109(2), *c* 17.019(2) Å, β = 111.850(10) $^\circ$, V 3198.4(8) Å³, Z 4, *d*_{calc} 1.164 g cm⁻³, P2₁/n space group. Intensities of 6156 reflections were measured in the reciprocal space ($2\Theta \leq 50$) using $\omega/2\Theta$ scanning. After exclusion of systematically cancelled reflections and averaging intensities of equivalent reflections, the working array of measured F²(hkl) and $\sigma(F^2)$ reflections contained 4660 independent reflections, 2745 of which with F²4 $\sigma(F^2)$. The structure was solved with the direct method and was refined by the full-matrix least-squares procedure with respect to F² in anisotropic approximation for non-hydrogen atoms using SHELXL-97 program.⁹ All hydrogen atoms were localized in the Fourier synthesis of the difference electronic density. The coordinates and isotropic thermal parameters were computed using the least-square "rider" model.⁹ In the final cycle of the full-matrix refinement, the absolute values of shifts of all variable 380 parameters of the structure of **6a** were less than 0.001 σ . The final refined parameters were: R₁ = 0.057, wr₂ = 0.13 for the reflections observed with I $\geq 2\sigma(i)$ and ; R₁ = 0.107, wr₂ = 0.15 for all measured reflections, GOF = 1.012. The maximum and minimum values of the difference electronic densities are 0.250 and -0.116258 e/ Å³. Atomic coordinates, full tables of bond lengths, bond angles and thermal parameters of **6a** have been deposited at the Cambridge Crystallographic Data Center (deposition numbers: CCDC 669813).

3,5-Di(*tert*-butyl)-2-[3-(7,8-dimethyl-4-morpholino-2-quinolyl)-2-quinoxaliny]-2,4-cyclopentadien-1-one (6a). A solution of 0.4 mmol of 1,7-di(*tert*-butyl)-3-(7,8-dimethyl-4-morpholino-2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-N-oxide (**3a**)³, 1.2 mmol of *o*-phenylenediamine and 50 mg of *p*-toluenesulfonic acid in 5 ml of toluene was heated for 4 h under reflux. The solution was cooled, the solvent evaporated and the residue was dissolved in 5 ml of chloroform. The chloroform solution was passed through an aluminium oxide column and the column was eluted with chloroform. A bright orange fraction was collected to give after evaporation of the solvent and recrystallization from 2-propanol **6a** in 67% yield. Orange crystals, m.p. 194-196 °C (from 2-propanol). IR-spectrum, ν , cm⁻¹: 1700, 1607, 1513, 1473, 1367, 1287, 1247. Mass spectrum, *m/z* (I_{rel}, %): 561 (45), 546 (35), 503 (65), 243 (15), 83 (30), 57 (40), 45 (100). ¹H NMR spectrum (CDCl₃): δ 0.89 (s, 9H, C(CH₃)₃), 1.21 (s, 9H, C(CH₃)₃), 2.46 (s, 3H, CH₃), 2.69 (s, 3H, CH₃), 3.20 - 3.45 (m, 4H), 3.90 - 4.10 (m, 4H), 6.80 (s, 1H), 7.28 (m, 1H), 7.79 (m, 4H), 8.13 (m, 1H), 8.21 (m, 1H). Anal. calcd. for C₃₆H₄₀N₄O₂ (560.73): C 77.11; H 7.19; N 9.99; O 5.71. Found: C 77.04; H 7.12; N 9.82.

3,5-Di(tert-butyl)-2-[3-(6,8-dimethyl-4-morpholino-2-quinolyl)-2-quinoxalinyl]-2,4-cyclopentadien-1-one (6b). It was obtained by coupling of 0.6 mmol 1,7-di(tert-butyl)-3-(6,8-dimethyl-4-morpholino-2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-N-oxide (**3b**), 1.2 mmol of *o*-phenylenediamine and 50 mg of *p*-toluenesulfonic acid as described for **6a**. Yield 74%, orange crystals, m.p. 182-184 °C. IR-spectrum, ν , cm^{-1} : 1700, 1580, 1513, 1473, 1380, 1233. Mass spectrum, m/z (I_{rel} , %): 561 (15), 546 (20), 503 (38), 243 (23), 77 (23), 57 (100), 41 (97). ^1H NMR spectrum (CDCl_3): δ 0.91 (s, 9H, $\text{C}(\text{CH}_3)_3$), 1.20 (s, 9H, $\text{C}(\text{CH}_3)_3$), 2.45 (s, 3H, CH_3), 2.70 (s, 3H, CH_3), 3.20 - 3.40 (m, 4H), 3.90 - 4.10 (m, 4H), 6.80 (s, 1H), 7.24 (m, 1H), 7.60 - 7.90 (m, 4H), 8.15 (m, 1H), 8.23 (m, 1H). Anal. calcd. for $\text{C}_{36}\text{H}_{40}\text{N}_4\text{O}_2$ (560.73): C 77.11; H 7.19; N 9.99; O 5.71. Found: C 77.02; H 7.15; N 9.73.

3,5-Di(tert-butyl)-2-[3-(8-methyl-4-morpholino-2-quinolyl)-2-quinoxalinyl]-2,4-cyclopentadien-1-one (6c). It was obtained by coupling 0.6 mmol 1,7-di(tert-butyl)-3-(8-methyl-4-morpholino-2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-N-oxide (**3c**), 1.2 mmol of *o*-phenylenediamine and 50 mg of *p*-toluenesulfonic acid as described for **6a**. Yield 61%, orange crystals, m.p. 184-186 °C. IR-spectrum, ν cm^{-1} : 1700, 1580, 1500, 1460, 1367. Mass spectrum, m/z (I_{rel} , %): 547 (13), 531 (17), 489 (23), 229 (17), 115 (28), 91 (17), 77 (25), 65 (15), 57 (96), 41 (100). ^1H NMR spectrum (CDCl_3): δ 0.90 (s, 9H, $\text{C}(\text{CH}_3)_3$), 1.19 (s, 9H, $\text{C}(\text{CH}_3)_3$), 2.74 (s, 3H, CH_3), 3.20 - 3.40 (m, 4H), 3.90- 4.10 (m, 4H), 6.80 (c, 1H), 7.30 - 7.50 (m, 2H), 7.70 - 7.90 (m, 4H), 8.12 (m, 1H), 8.21 (m, 1H). Anal. calcd. for $\text{C}_{35}\text{H}_{38}\text{N}_4\text{O}_2$ (546.71): C 76.89; H 7.01; N 10.25; O 5.85. Found: C 76.62; H 7.04; N 10.02.

3,5-Di(tert-butyl)-2-[3-(7,8-dimethyl-4-morpholino-2-quinolyl)-5,8-dimethyl-2-quinoxalinyl]-2,4-cyclopentadien-1-one (6d). It was obtained by coupling 0.6 mmol 1,7-di(tert-butyl)-3-(7,8-dimethyl-4-morpholino-2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-N-oxide (**3a**), 1.2 mmol of 3,6-dimethyl-*o*-phenylenediamine and 50 mg of *p*-toluenesulfonic acid as described for **6a**. Yield 57%, orange crystals, mp 253-255 °C. Mass spectrum, m/z (I_{rel} , %): 589 (3), 532 (10), 243 (18), 103 (12), 91 (15), 77 (25), 57 (100), 41 (89). ^1H NMR spectrum (CDCl_3): δ 0.82 (s, 9H, $\text{C}(\text{CH}_3)_3$), 1.36 (s, 9H, $\text{C}(\text{CH}_3)_3$), 2.45 (s, 3H, CH_3), 2.70 (s, 3H, CH_3), 2.79 (s, 3H, CH_3), 2.86 (s, 3H, CH_3), 3.20 - 3.40 (m, 4H), 3.90 - 4.10 (m, 4H), 6.89 (s, 1H), 7.40 - 7.60 (m, 2H), 7.70 - 7.85 (m, 2H), 8.12 (m, 1H), 8.08 (s, 1H). Anal. calcd. for $\text{C}_{38}\text{H}_{44}\text{N}_4\text{O}_2$ (588.79): C 77.52; H 7.53; N 9.52; O 5.43. Found: C 77.38; H 7.44; N 9.42.

3,5-Di(tert-butyl)-2-[3-(8-methyl-4-morpholino-2-quinolyl)-5,8-dimethyl-2-quinoxalinyl]-2,4-cyclopentadien-1-one (6e). It was obtained by coupling 0.4 mmol 1,7-di(tert-butyl)-3-(8-methyl-4-morpholino-2-quinolyl)-2-azabicyclo[3.3.0]octa-2,7-dien-4,6-dione-N-oxide (**3c**), 0.8 mmol of 3,6-dimethyl-*o*-phenylenediamine and 50 mg of *p*-toluenesulfonic acid as described for **6a**. Yield 65%, orange crystals, m.p. 247-249 °C. IR-spectrum, ν , cm^{-1} : 1700, 1580, 1527, 1500, 1460, 1233. Mass spectrum, m/z (I_{rel} , %): 575 (10), 560 (11), 517 (27), 229 (23), 115 (19), 91 (16), 77 (25), 65 (10), 57 (100), 41 (87). ^1H NMR spectrum (CDCl_3): δ 0.84 [s, 9H, $\text{C}(\text{CH}_3)_3$], 1.35 [s, 9H, $\text{C}(\text{CH}_3)_3$], 2.75 (s, 3H, CH_3), 2.79 (s, 3H, CH_3), 2.86 (s, 3H, CH_3), 3.20 – 3.40 (m, 4H), 3.90 – 4.10 (m, 4H), 6.88 (s, 1H), 7.30 - 7.50 (m, 4H), 7.88 (m, 1H), 8.12 (s, 1H). Anal.

calcd. for C₃₇H₄₂N₄O₂ (574.76): C 77.32; H 7.37; N 9.75; O 5.57. Found: C 77.16; H 7.24; N 9.48.

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