# Heterocyclic synthesis with activated nitriles : an expeditus synthetic approach to polyfunctionally substituted pyrroles, heterocyclopyrimidines and coumarins

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### Abstract

The applicability and synthetic potency of the new reagent N-(2,2-dicyanoethenyl)aminoacetonitrile (3) to develop an expeditious convenient synthetic route of unique polyfunctionally substituted pyrroles, heterocyclopyrimidines and 2*H*-1-benzopyran-2-ones is reported. Chemical and spectroscopic evidence for the structures of the newly synthesized compounds are described.

**Keywords** : Nitriles, pyrroles, pyrimidines, coumarins

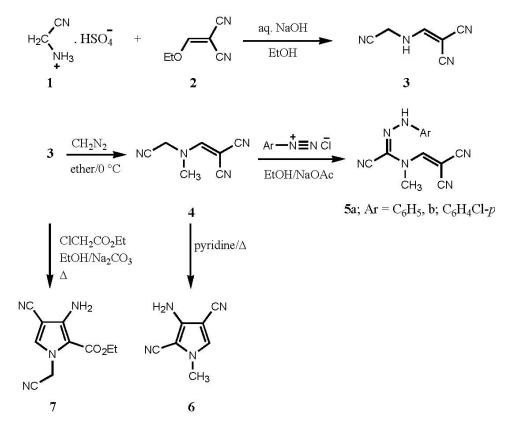
α, β-π-Deficient nitrile reagents are highly reactive and are extensively utilized as unique synthons for the construction of a variety of unique heterocyclic ring systems.<sup>1–5</sup> In continuation with our medicinal chemistry program directed towards the development of new procedures for the synthesis of azoles, azines and their condensed derivatives utilizing readily obtainable polyfunctional nitriles,<sup>6–10</sup> we report herein, an easy, facile and new synthetic methodology for the synthesis of *N*-(2,2-dicyanoethenyl)aminoacetonitrile (**3**). The latter proved to be a new key precursor for the synthesis of some polyfunctionally substituted pyrroles, thiazolo[1,2-*a*]pyrimidines, and coumarins. The importance of those compounds is due to their diverse potential biological and physiological expected broad spectrum.<sup>11–14</sup>

# **Results and Discussion**

The key precursor N-(2,2-dicyanoethenyl)aminoacetonitrile (3) was prepared, in a facile high yield reaction, upon treatment of aminoacetonitrile hydrogen sulphate (1) with an equimolar amount of ethyl 2-cyano-3-ethoxypropenonitrile (2) in an aqueous alcoholic NaOH solution at

room temperature. Assignment of structure **3** for the reaction product was based on its correct elemental analyses and compatible spectroscopic data. Thus, its mass spectrum revealed a molecular ion peak at m/z (%) = 132 (18%) corresponding to the molecular formula C<sub>6</sub>H<sub>4</sub>N<sub>4</sub>. Its IR spectrum showed absorption peaks at v 3340 (NH), and 2220, 2216, 2208 cm<sup>-1</sup> (3CN). Its <sup>1</sup>H-NMR spectrum (DMSO-*d*<sub>6</sub>) showed a singlet signal (2H) at  $\delta$  4.12 ppm assigned for the CH<sub>2</sub> protons, a singlet at  $\delta$  7.32 ppm assigned for the CH proton and a broad singal at  $\delta$  9.11 ppm assigned for the NH proton which underwent a facile hydrogen deuterium exchange and disappeared upon addition of D<sub>2</sub>O to the NMR sample. The structure of **3** was further confirmed on the basis of its chemical behaviour towards different chemical reagents.

Compound **3** reacted with an equimolar amount of diazomethane in dry ether at ice bath to yield the corresponding *N*-methyl derivative **4**. The active methylene group in compound **4** underwent an electrophilic substitution upon coupling with equimolar amounts of aryldiazonium chloride salts to yield the corresponding hydrazone coupling products **5a,b**. Compound **5** proved to exist predominantly in the hydrazone form rather than the azo form on the basis of <sup>1</sup>H-NMR data. Thus, e.g. **5a** revealed, besides the expected signals, the presence of a singlet signal exchangeable with D<sub>2</sub>O at  $\delta$  8.21 ppm attributed to the hydrazone NH function. Furthermore, its UV spectrum showed absorption maxima at  $\lambda$  420 ( $\epsilon$  = 35 280) and 330 nm ( $\epsilon$ = 23 470) in accordance with those for hydrazo functions which are reported to exhibit strong absorption at wavelengths higher than 310 nm.<sup>15,16</sup>

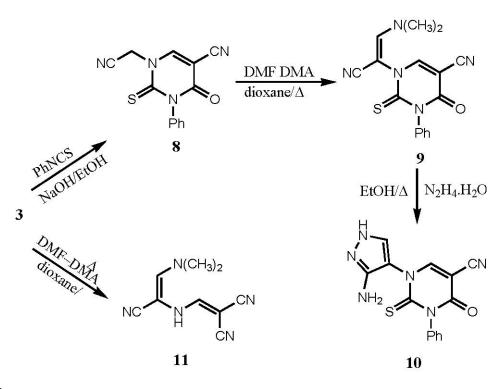


#### Scheme 1

Compound **4** underwent an intramolecular heterocyclization, upon boiling under reflux in dry pyridine, to afford 3-amino-2,4-dicyano-1-methylpyrrole (**6**). Compound **3** readily reacted with an equimolar amount of ethyl chloroacetate in aqueous  $Na_2CO_3$  solution, under reflux, to yield ethyl 3-amino-4-cyano-1-cyanomethylpyrrole-2-carboxylate (**7**) (Scheme 1). Formation of **7** is assumed to proceed *via* an initial N-alkylation and a subsequent intramolecular heterocyclization *via* a Michael-type nucleophilic addition of the CH<sub>2</sub> protons to the cyano function.

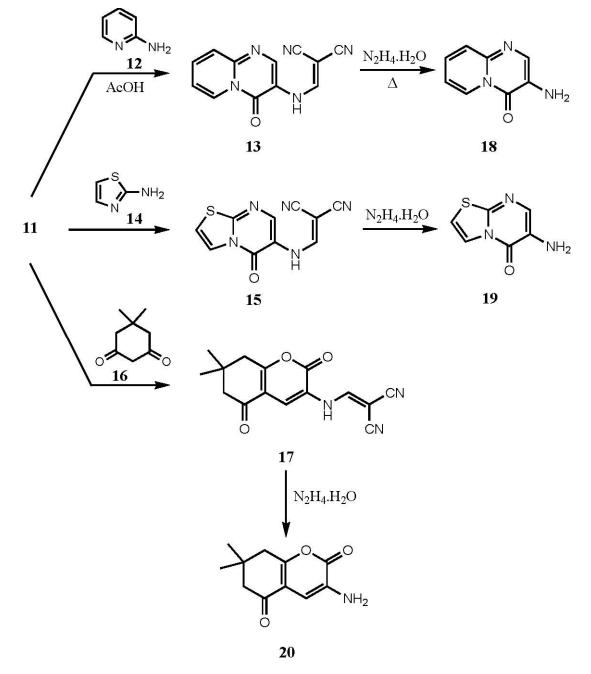
The base-promoted nucleophilic addition of phenyl isothiocyanate to **3** has been achieved, upon boiling under reflux in an aqueous ethanolic NaOH solution, to yield the corresponding 1cyanomethyl-4-oxo-2-thioxopyrimidine derivative **8** (Scheme 2). The reaction apparently involves an intramolecular heterocyclization *via* the addition of the amine nitrogen of **3** to the carbon of isothiocyanate function and a subsequent alkaline hydrolysis of the imino function under the reaction conditions.17 Reaction of **8** with an equimolar amount of *N*,*N*dimethylformamide dimethylacetal (DMF-DMA) in dry dioxane, under reflux, furnished 5cyano-1-[1'-cyano-2'-dimethylaminoethenyl)-4-oxo-3-phenyl-1,2,3,4-tetra-hydro-2-

thioxopyrimidine (9) in an acceptable yield. Treatment of 9 with an equimolar amount of hydrazine hydrate in refluxing ethanol furnished exclusively a single product that could be formulated as 1-[3'-amino-pyrazol-4'-yl]-5-cyano-4-oxo-3-phenyl-1,2,3,4-tetrahydro-2-thioxopyri-midine (10). Characterization of 10 is consistent with its elemental analysis and spectral data. However, compound 10 was an unstable yellow product that decomposed automatically at 25 °C within two days, therefore, it should be stored in a refrigerator. When compound 3 was treated with an equimolar amount of *N*,*N*-dimethylformamide dimethylacetal (DMF-DMA) in dry dioxane, under reflux, the corresponding 2,2-dimethylaminoethenyl derivative 11 was isolated. Elucidation of the proposed structure 11 was based on its correct elemental analyses and compatible spectroscopic data.



#### Scheme 2

Next, it was of interest to explore the scope, limitations and generality of **11** as a precursor for the synthesis of some difficult to access polyfunctionally substituted fused pyrimidine or benzopyranone derivatives for which we might expect a wide spectrum of bioresponses. Thus, reaction of **11** with equimolar amounts of *N*-nucleophiles; namely, 2-aminopyridine or 2-amino-thiazole, upon reflux in glacial acetic acid, afforded the corresponding 4H-pyrido[1,2-a]pyrimidin-4-one **13** and 5H-thiazolo[1,2-a]pyrimidin-5-one **15** derivatives, respectively, in reasonable yields (Scheme 3). The identity of the product was established on the basis of elemental analyses and spectral background in each case.



#### Scheme 3

Analogously, compound **11** reacted with *C*-nucleophiles e.g. dimedone, under the same experimental conditions to yield the corresponding 5,6,7,8-tetrahydro-2*H*-1-benzopyran-2-one derivative **17**.

The free amino heterocycles **18-20** could be achieved on treatment of the appropriate 2,2dicyanoethenylamino heterocycles **13**, **15**, or **17** with hydrazine hydrate in ethanol at reflux temperature, respectively.

In conclusion, the results presented in this article, indirectly extend and broaden the

knowledge in the area of activated nitriles and demonstrate a general applicable methodology for the construction of a wide variety of polyfunctionally substituted annelated pyrimidines and 2H-1-benzopyranones of expected wide spectrum of potential bioresponses. Work along the expansion of such synthetic approach is now in progress.

### **Experimental Section**

General Procedures. Melting points are uncorrected. IR spectra were recorded (KBr disc) on a Pye Unicam SP-1000 spectrophotometer. <sup>1</sup>H-NMR spectra were obtained on a Varian Gemini 200 MHz spectrophoto-meter using *TMS* as an internal reference. Chemical shifts are expressed as  $\delta$  (ppm). Mass spectra were recorded on a GCMS-QP 1000 Ex spectra mass spectrometer operating at 70 eV. Microanalytical data were performed by the Microanalytical Data Unit, Cairo University.

*N*-(2,2-Dicyanoethenyl)aminoacetonitrile (3). To a solution of aminoacetonitrile hydrogen sulphate (1) (100 mmol, 15.4 g) in water (60 mL) containing sodium hydroxide (100 mmol, 4.0 g), a solution of ethyl 2-cyano-3-ethoxypropenoate (2) (100 mmol, 12.2 g) in EtOH (50 mL) was added portionwise. The reaction mixture was stirred for 2 h and left aside overnight at room temperature. The solid product that separated was collected by filtration and crystallized from ethanol to give 10.04 g (76%); mp 45–7 °C; IR (KBr) 3340 (NH), 2220, 2216, 2208 cm<sup>-1</sup> (3CN); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  4.12 (s, 2H, CH<sub>2</sub>), 7.32 (s, 1H, CH), 9.11 (br s, 1H, NH, exchangeable); *m/z* (%) 132 (M<sup>+</sup>, 18%); Anal. Calcd for C<sub>6</sub>H<sub>4</sub>N<sub>4</sub>: C, 54.5; H, 3.05; N, 42.4. Found: C, 54.4; H, 3.0; N, 42.2.

*N*-(2,2-Dicyanoethenyl)-*N*-methylaminoacetonitrile (4). To a solution of 3 (20 mmol, 2.64 g) in dry ether (50 mL), diazomethane (20 mmol, 0.84 g) was added. The reaction mixture was stirred at ice-bath temperature 0–5 °C for 2 h, then left aside overnight at 0 °C. The mixture was evaporated under reduced pressure at 25 °C, whereby the residue that precipitated was collected by filtration and crystallized from ether. Yield 2.07 g (71%); mp 49 °C; IR (KBr) 2220, 2214, 2210 cm<sup>-1</sup> (3CN); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  2.12 (s, 3H, CH<sub>3</sub>), 4.08 (s, 2H, CH<sub>2</sub>), 7.18 (s, 1H, CH); Anal. Calcd for C<sub>7</sub>H<sub>6</sub>N<sub>4</sub>: C, 57.5; H, 4.1; N, 38.3. Found: C, 57.4; H, 4.1; N, 38.2.

**Arylhydrazono-***N***-(2,2-dicyanoethenyl)**-*N***-methylaminoacetonitriles 5a,b**. (*General Procedure*). To a stirred solution of **4** (5 mmol, 0.73 g) in ethanol (50 mL) containing NaOAc (1.0 g), the appropriate aryldiazonium chloride (5 mmol) [prepared by adding NaNO<sub>2</sub> (5 mmol, 0.35 g) to the appropriate primary aromatic amine (5 mmol) in concentrated HCl (2 mL) at 0–5 °C while stirring] was added dropwise while cooling at 0–5 °C and stirring. The reaction mixture was left aside at room temperature for 3 h, whereby the solid product that separated was collected by filtration, dried and crystallized from the appropriate solvent.

*N*-(2,2-Dicyanoethenyl)-*N*-methylphenylhydrazonoaminoacetonitriles (5a). 0.70 g (56%); mp 105 °C (EtOH); IR (KBr) 2222, 2218, 2209 cm<sup>-1</sup> (3CN); 1H-NMR (DMSO- $d_6$ )  $\delta$  2.05 (s, 3H,

CH<sub>3</sub>), 7.16 (s, 1H, CH), 7.32-7.39 (m, 5H, aromatic protons), 8.21 (s, 1H, NH, exchangeable); m/z (%) 250 (M<sup>+</sup>, 22%); Anal. Calcd for C<sub>13</sub>H<sub>10</sub>N<sub>6</sub>: C, 62.4; H, 4.0; N, 33.6. Found: C, 62.2; H, 4.0; N, 33.5.

**p-Chlorophenylhydrazono**-*N*-(**2'**,**2'-Dicyanoethenyl**)-*N*-methylamino-acetonitriles (**5b**). 0.88 g (62%); mp 121 °C (EtOH); IR (KBr) 3342 (NH), 2220, 2218, 2210 cm<sup>-1</sup> (3CN); <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$  2.08 (s, 3H, CH<sub>3</sub>), 7.14 (s, 1H, CH), 7.30-7.36 (m, 4H, aromatic protons), 8.52 (s, 1H, NH, exchangeable); Anal. Calcd for C<sub>13</sub>H<sub>9</sub>ClN<sub>6</sub>: C, 54.8; H, 3.2; Cl, 12.45; N, 29.5. Found: C, 54.8; H, 3.0; Cl, 12.3; N, 29.5.

**3-Amino-2,4-dicyano-1-methylpyrrole** (6). A solution of **4** (5 mmol, 0.73 g) in dry pyridine (30 mL) was refluxed for 3 h. The reaction mixture was cooled at room temperatures triturated with water, whereby the resulted solid product was collected by filtration, dried and crystallized from AcOH. Yield 0.39 g (54%); mp 152–4 °C (EtOH); IR (KBr) 3450-3380 (NH<sub>2</sub>), 2218, 2212 cm<sup>-1</sup> (2 CN); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  2.31 (s, 3H, CH<sub>3</sub>), 6.33 (s, 1H, pyrrole 5-H), 10.71 (br s, 2H, NH<sub>2</sub>, exchangeable); Anal. Calcd for C<sub>7</sub>H<sub>6</sub>N<sub>4</sub>: C, 57.5; H, 4.1; N, 38.3. Found: C, 57.4; H, 4.0; N, 38.2.

Ethyl 3-amino-4-cyano-1-cyanomethylpyrrole-2-carboxylate (7). To a warm solution of 3 (5 mmol, 0.66 g) in ethanol (30 mL), ethyl chloroacetate (5 mmol, 0.61 g) in an aqueous Na<sub>2</sub>CO<sub>3</sub> solution [(5 mmol, 0.28 g in H<sub>2</sub>O (10 mL)] was added. The reaction mixture was refluxed for 2 h, left to cool at room temperature, poured onto cold water and neutralized with dilute HCl. The solid product separated was filtered off, dried and crystallized from AcOH. Yield 0.60 g (55%); mp 142 °C (EtOH); IR (KBr) 3445-3380 (NH<sub>2</sub>), 2218, 2210 cm<sup>-1</sup> (2CN); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  1.10 (t, 3H, CH<sub>3</sub>, *J* = 7.2 Hz), 4.00 (q, 2H, CH<sub>2</sub>, *J* = 7.2 Hz), 4.12 (s, 2H, CH<sub>2</sub>), 6.31 (s, 1H, pyrrole 5-H), 10.62 (br s, 2H, NH<sub>2</sub>, exchangeable); *m/z* (%) 218 (M<sup>+</sup>, 18%). Anal. Calcd for C<sub>10</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub>: C, 55.0; H, 4.6; N, 25.7. Found: C, 55.0; H, 4.5; N, 25.6.

**1-Cyanomethyl-4-oxo-3-phenyl-1,2,3,4-tetrahydro-2-thioxopyrimidine** (**8**). To a solution of **3** (5 mmol, 0.66 g) in ethanol (30 mL), phenyl isothiocyanate (5 mmol, 0.67 g) in aqueous solution of NaOH (5 mmol, 0.20 g in 10 mL H<sub>2</sub>O) was added. The reaction mixture was refluxed for 3 h, left to cool at room temperature, poured onto cold water and neutralized with dilute HCl. The formed precipitate was filtered off, dried and crystallized from AcOH. Yield 0.86 g (64%); mp 163 °C (EtOH); IR (KBr) 2220, 2216 (2CN), 1690 (CO) cm<sup>-1</sup>; 1H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  4.08 (s, 2H, CH<sub>2</sub>), 6.18 (s, 1H, pyrimidine 4-H), 7.28-7.34 (m, 5H, aromatic protons); Anal. Calcd for C<sub>13</sub>H<sub>8</sub>N<sub>4</sub>OS: C, 58.2; H, 3.0; N, 20.9; S, 11.95. Found: C, 58.0; H, 2.9; N, 20.7; S, 11.8.

5-Cyano-1-[1'-cyano-2'-dimethylaminoethenyl)-4-oxo-3-phenyl-1,2,3,4-tetrahydro-2-

**thioxopyrimidine** (9). To a solution of **8** (5 mmol, 1.34 g) in dry dioxane (30 mL), dimethylformamide dimethylacetal (DMF-DMA) (5 mmol, 0.6 g) was added. The reaction mixture was refluxed for 3 h, evaporated *in vacuo* and triturated with ethanol. The resulting solid product was collected by filtration, dried and crystallized from dioxane. Yield 0.79 g (49%); mp 175 °C (EtOH); IR (KBr) 2218, 2215 (2CN), 1686 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  2.86 (s, 6H, 2CH<sub>3</sub>), 6.12 (s, 1H, pyrimidine 4-H), 7.26-7.31 (m, 5H, aromatic protons), 8.22 (s, 1H, CH); Anal. Calcd for C<sub>16</sub>H<sub>13</sub>N<sub>5</sub>OS: C, 59.4; H, 4.05; N, 21.65; S, 9.9. Found: C, 59.4; H, 3.8; N, 21.4;

### S, 9.8.

1-[3'-Aminopyrazol-4'-yl]-5-cyano-4-oxo-3-phenyl-1,2,3,4-tetrahydro-2-thioxopyrimidine

(10). A mixture of 9 (3 mmol, 0.97 g) and hydrazine hydrate (3 mmol, 0.97 g) in ethanol (30 mL) was boiled under reflux for 30 min, and then left aside at room temperature overnight. The mixture was poured onto cold water, whereby the solid product was filtered off and crystallized from EtOH. Yield 0.52 g (56%); mp 138 °C (EtOH); IR (KBr) 3445-3210 (NH and NH<sub>2</sub>), 2218 (CN), 1690 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  6.05 (s, 1H, pyrazole 5-H), 6.22 (s, 1H, pyrimidine 4-H), 7.26-7.31 (m, 5H, aromatic protons), 8.46 (s, 1H, NH, exchangeable), 9.62 (br s, 2H, NH<sub>2</sub>, exchangeable); Anal. Calcd for C<sub>14</sub>H<sub>10</sub>N<sub>6</sub>OS: C, 54.2; H, 3.25; N, 27.1; S, 10.3. Found: C, 54.0; H, 3.2; N, 26.9; S, 10.2.

*N*-(2,2-Dicyanoethenyl)-2,2-dimethylaminoethenyl)aminoacetonitrile (11). To a solution of 3 (10 mmol, 1.32 g) in dry dioxane (30 mL), dimethylformamide dimethylacetal (DMF-DMA) (10 mmol, 1.2 g) was added. The reaction mixture was refluxed for 3 h, evaporated *in vacuo* and triturated with ethanol. The resulting solid product was filtered, dried and crystallized from dioxane. Yield 1.01 g (54%); mp 135 °C (EtOH); IR (KBr) 3345 (NH), 2218, 2216, 2210 cm<sup>-1</sup> (3 CN); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  2.81 (s, 6H, 2CH<sub>3</sub>), 7.15 (d, 1H, CH), 8.16 (br s, 1H, NH, exchangeable), 8.22 (s, 1H, CH); Anal. Calcd for C<sub>9</sub>H<sub>9</sub>N<sub>5</sub>: C, 57.7; H, 4.8; N, 37.4. Found: C, 57.6; H, 4.8; N, 37.3.

**Reaction of 11 with heteroaryl amines 12, 14 and with C-nucleophile 16**. (*General procedure*). To a solution of **11** (4 mmol, 0.75 g) in glacial AcOH (30 mL), the appropriate heteroaryl amine or C-nucleophile (4 mmol) was added. The reaction mixture was refluxed for 3 h and then left aside at room temperature overnight. The mixture was evaporated under reduced pressure, the residue was triturated with ethanol, whereby the solid precipitated so-formed was collected by filtration, dried and crystallized from the appropriate solvent.

**3-(2,2-Dicyanoethenyl)amino-4***H***-pyrido[1,2-***a***]pyrimidin-4-one (13). Yield 0.39 g (41%); mp 171 °C (EtOH); IR (KBr) 3420, 3415 (NH), 2220, 2218 (2CN), 1695 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-***d***<sub>6</sub>) \delta 7.21-7.32 (m, 6H, aromatic protons + CH)), 8.95 (s, 1H, NH, exchangeable); m/z (%) 237 (M<sup>+</sup>, 16%); Anal. Calcd for C<sub>12</sub>H<sub>7</sub>N<sub>5</sub>O: C, 60.8; H, 3.0; N, 29.5. Found: C, 60.6; H, 2.9; N, 29.3.** 

**6-(2,2-Dicyanoethenyl)amino-5***H***-thiazolo[1,2-***a***]pyrimidin-5-one (15). Yield 0.38 g (49%); mp 193 °C (EtOH); IR (KBr) 3435 (NH), 2220, 2216 (2 CN), 1690 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-***d***<sub>6</sub>) δ 7.08 (s, 1H, pyrimidine H-4), 7.16 (d, 2H, thiazole H-4, H-5), 7.28 (s, 1H, CH), 8.92 (s, 1H, NH, exchangeable); Anal. Calcd for C<sub>10</sub>H<sub>5</sub>N<sub>5</sub>OS: C, 49.4; H, 2.1; N, 28.8; S, 13.2. Found: C, 49.3; H, 1.9; N, 28.6; S, 13.0.** 

**3-(2,2-Dicyanoethenyl)amino-7,7-dimethyl-5-oxo-5,6,7,8-tetrahydro-2***H***-1-benzopyran-2one (17). Yield 0.49 g (43%); mp 210 °C (EtOH); IR (KBr) 3435 (NH), 2220, 2216 (2CN), 1690, 1682 cm<sup>-1</sup> (2 CO); <sup>1</sup>H-NMR (DMSO-d\_6) \delta 1.24 (s, 6H, 2CH<sub>3</sub>), 2.48 (s, 2H, CH<sub>2</sub>), 2.73 (s, 2H, CH<sub>2</sub>), 7.58 (s, 1H, H-4), 7.72 (d, 1H, CHNH), 10.52 (d, 1H, NH, exchangeable); Anal. Calcd for C<sub>15</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub>: C, 63.6, H, 4.6; N, 14.8. Found: C, 63.5; H, 4.6; N, 14.7.**  **Preparation of the free aminoheterocycles 18, 19 and 20** (*General procedure*). A mixture of the N-(2,2-dicyanoethenyl)amino heterocyclic compound 13, 15 or 17 (2 mmol) and hydrazine hydrate (2 mmol, 0.1 g) in ethanol (15 mL) was refluxed for 1 h, and then left aside at room temperature overnight. The precipitated resulted was filtered off and crystallized from the appropriate solvent.

**3-Amino-4H-pyrido**[1,2-*a*]**pyrimidin-4-one** (18). Yield 0.17 g (52%); mp 178-180 °C (EtOH) (Lit.<sup>18</sup> mp 180-2 °C); IR (KBr) 3450-3375 (NH<sub>2</sub>), 1690 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  5.57 (br s, 2H, NH<sub>2</sub>, exchangeable), 7.21-7.38 (m, 5H, aromatic protons); *m/z* (%) 161 (M<sup>+</sup>, 14%); Anal. Calcd for C<sub>8</sub>H<sub>7</sub>N<sub>3</sub>O: C, 59.6; H, 4.4; N, 26.1. Found: C, 59.5; H, 4.3; N, 25.9.

**6-Amino-5H-thiazolo[3,2-***a***]pyrimidin-5-one (19)**. Yield 0.22 g (66%); mp 170-2 °C (EtOH); IR (KBr) 3454-3375 (NH<sub>2</sub>), 1688 cm<sup>-1</sup> (CO); <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$  5.14 (br s, 2H, NH<sub>2</sub>, exchangeable), 7.15 (s, 1H, pyrimidine 4-H), 7.23 (d, 2H, thiazole H-4, H-5); Anal. Calcd for C<sub>6</sub>H<sub>5</sub>N<sub>3</sub>OS: C, 43.1; H, 3.0; N, 25.1; S, 19.2. Found: C, 43.0; H, 2.8; N, 24.9; S, 19.0.

**3-Amino-7,7-dimethyl-5-oxo-5,6,7,8-tetrahydro-2H-1-benzopyran-2-one** (**20**). Yield 0.20 g (49%); mp 149 °C (EtOH); IR (KBr) 3450-3364 (NH<sub>2</sub>), 1695, 1688 cm<sup>-1</sup> (2 CO); <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$  1.26 (s, 6H, 2CH<sub>3</sub>), 2.44 (s, 2H, CH<sub>2</sub>), 2.63 (s, 2H, CH<sub>2</sub>), 5.35 (br s, 2H, NH<sub>2</sub>, exchangeable), 7.40 (s, 1H, H-4); Anal. Calcd for C<sub>11</sub>H<sub>13</sub>NO<sub>3</sub>: C, 63.75; H, 6.3; N, 6.8. Found: C, 63.6; H, 6.3; N, 6.6.

# References

- 1. Erian, A. W. Chem. Rev. 1993, 93, 1991 and references cited therein.
- 2. Erian, A. W. J. Heterocycl. Chem. 2001, 38, 793 and references cited therein.
- 3. Erian, A. W.; Sherif, S. M. Tetrahedron 1999, 55, 7957 and references cited therein.
- 4. Sherif, S. M.; Erian, A. W. *Heterocycles* **1996**, *43*, 1085 and references cited therein.
- 5. Elnagdi, M. H.; Sherif, S. M.; Mohareb, R. M. *Heterocycles* **1987**, *26*, 496 and references cited therein.
- 6. Erian, A. W.; Issac, Y. A.; Sherif, S. M.; Mahmoud, F. F. J. Chem. Soc., Perkin Trans. 1 2000, 3686.
- 7. Erian, A. W.; Issac, Y. A.; Sherif, S. M. Z. Naturforsch. 2000, 55b, 127.
- 8. Erian, A. W.; Araki, V. F.; Aziz, S. I.; Sherif, S. M. Monatsh. Chem. 1999, 130, 661.
- 9. Sherif, S. M.; Youssef, M. M.; Mobark, K. M.; Abdel-Fattah, A. M. *Tetrahedron* **1993**, *49*, 9561.
- 10. Erian, A. W.; Sherif, S. M.; Alasser, A. A.; Elkholy, Y. M. Tetrahedron 1994, 50, 1877.
- 11. Ram, V. J.; Kushwaha, D. S.; Mishra, L. Indian J. Chem. 1989, 28B, 242.
- 12. Suguira, K.; Schind, A. F.; Schimd, M. M.; Brown F. G. *Cancer Chemother. Rep.* **1971**, *3*, 231.
- 13. Shishoo, C. J.; Jain, K. S. J. Heterocycl. Chem. 1992, 29, 883.

- 14. Elslager, E. F.; Hess, C.; Johnson, J.; Ortwine, D.; Chien, V.; Werbel, L. M. J. Med. Chem. 1981, 24, 127.
- 15. Yao, H. C.; Resnick, P. J. J. Am. Chem. Soc. 1962, 84, 3514.
- 16. Yao, H. C. J. Org. Chem. 1964, 29, 2959.
- 17. Methcohn, O.; Tarnowski, B. Synthesis 1978, 56.
- 18. Selic, L.; Grdadolink, S. G.; Stanovnik, B. Heterocycles 1998, 49, 133.