# An unexpected ring formation in morphine chemistry

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Dedicated to Prof. Fritz Sauter on the occasion of his 70<sup>th</sup> birthday (received 23 Dec 00; accepted 29 Oct 01; published on the web 06 Nov 01)

#### **Abstract**

Reaction of (7*R*)-*N*-benzyl-*N*-nor-7-bromoneopinone dimethyl acetal (2) with primary or secondary amines (RNH2, R1R2NH) resulted in 9-amino-substituted hasubanan derivatives; in particular, modification of the 9-(2-hydroxyethyl)amino derivative in subsequent steps gave products 8 with a 9-(oxazolidin-2'-one-3'-yl) substituent. <sup>1</sup>H and <sup>13</sup>C NMR spectra did not provide definitive evidence for the supposed structure and reaction mechanism, but X-ray analyses of compounds 8a and 8d confirmed both structures.

**Keywords:** Morphine derivatives, neopine derivatives, hasubanan, ring formation, 9-(oxazolidin-2'-one-3'-vl)-substitution, X-ray structure analysis

#### Introduction

It is known that the position of basic nitrogen in the morphine scaffold as well as the type of its substituents are of great importance with regard to the interaction with the anionic site of the receptor. The goal of our approach was to introduce a second basic group in the vicinity to the nitrogen attached to C-9. The subsequent step should turn the original nitrogen atom into a non-basic group, incapable of interaction with the anionic site of the receptor. In this way the scaffold nitrogen would be changed to inefficiency whereas the newly inserted one could be capable of pharmacologic interactions.

Two more points of consideration should be mentioned: 1. A second nitrogen offers the advantage to insert an intramolecular bridge. 2. The synthesis of dimers by inserting suitable bivalent reagents should be possible; such compounds are known to possess higher

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antinociceptive potency and receptor selectivity by entropic reasons.1

#### Scheme 1

We started as follows (Scheme 1): Reaction of the known 14,17-cyclonorcodeinone dimethyl acetal 1² with benzyl bromide yielded quantitatively the neopinone derivative 2.³ Generally, hydrolysis, methanolysis, acetolysis, *i.e.* nucleophilic substitution of 7-bromoneopinone or 14-bromocodeinone acetal derivatives are assumed to occur via an ionic intermediate (a quaternary aziridinium ion derived from 1), which results in a mixture of 7-, 14- and 9-substituted derivatives. ⁴ Therefore, the straightforward reaction with primary and secondary amines furnishing single, defined products was a surprise at first sight. Inspection of the NMR spectra of the product eliminated structure 5, but it was not possible to discriminate unambiguously between amine substitution at C-9 (as in 3) or C-14 (as in 4) by any routine analytical method. Only the ¹H NMR spectrum of derivative 3d, obtained later, gave evidence in support of the proposed structure based on the paramagnetic shift of the H-9 signal as a consequence of carbamate formation. We expected that synthetic steps to be carried out at a later stage and aiming at cyclisation by bridging both nitrogen atoms would support this indication, but at last the X-ray analyses of 8a and 8d confirmed the substitution at C-9.

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#### **Results and Discussion**

In regard to our experience with codeinone derivatives, bridge building between both nitrogen atoms at C-9 and C-14 should be possible.<sup>5</sup> In **3c** and **4c** ( $R_1 = H$ ,  $R_2 = CH_2CH_2OH$ ) the hydroxyl group is expected to become a good nucleofuge upon conversion into the tosylate such as **3e** or **4e**, thus facilitating ring closure with the other nitrogen (N-17 of the original morphine scaffold) to form ammonium salt **6** or **7**, respectively (Scheme 2), which in turn, could afford a dibasic compound after detachment of the benzyl groups in a subsequent step.

#### Scheme 2

At this stage we engaged in some mechanistic considerations to evaluate the possibility of an alternative ring formation from **3e** or **4e** (bond lengths, steric energies). It must be argued that at first sight there seemed to be no energy preference for ring formation taking place from a newly inserted *N*-substituent at C-9 to the rearranged N-17, now located at C-14 (in **3**), or alternatively from the alkylamino substituent at C-14 to N-17 bonded to C-9 as in the original scaffold (in **4**). In fact, computation (MM2 force field) of the energies of the alternative bases derived from structures **6** and **7** gave almost identical results (14.7 vs. 14.9 kJmol-1), whereas the quaternary ammonium salts **6** and **7** revealed a rather low steric energy of 16.4 kJmol-1 for **6**, *vs*. 18.6 kJmol-1 for **7**, obviously due to the more crowded topology caused by substitution with the bulkier Z-group in the vicinity of the quaternary C-14. Since the reaction did not work efficiently with the unprotected secondary amine **3c** or **4c**, the benzyloxycarbonyl group (Z-group) was introduced as R<sub>1</sub> to give **3d** or **4d**. The subsequent tosylation step to form **3e** or **4e** was carried out without

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isolation of the resulting intermediate and was expected to facilitate the desired intramolecular reaction by increasing the electrophilicity of the terminal methylene group (Scheme 2).

Surprisingly, we did not obtain the quaternary ammonium salt 6 or 7, but rather a product in high yield and with concomitant loss of *one* benzyl group. Later this product was identified as the oxazolidinone derivative 8a (vide infra). At first, we were not aware of this and believed that a diazine moiety had been formed, and keeping in mind the loss of one benzyl group we set out to distinguish between a product structure derived from either 6 or 7.

It is noteworthy to mention that the decisive condition for this unique reaction leading to **8a** is the presence of dimethylaminopyridine (DMAP) in the tosylation step which triggered the entire reaction sequence: nucleophilic attack of DMAP at the *O*-benzyl methylene group induced CH2–O-bond cleavage and ring closure by the carboxyl oxygen at the tosyloxymethylene group with displacement of the tosylate anion. The benzyl group of the Z-group was consumed by DMAP affording the *N*-benzyl-4-(dimethyl-amino)pyridinium ion, which was precipitated as the perchlorate salt and analyzed after crystallisation. Without the assistance of DMAP no reaction was detectable.

_8	R <sup>1</sup>	R <sup>2</sup>
а	CH <sub>2</sub> Ph	$(OCH_3)_2$
b	Н	$(OCH_3)_2$
С	CH <sub>3</sub>	$(OCH_3)_2$
d	CH <sub>2</sub> CH=CH <sub>2</sub>	$(OCH_3)_2$
е	CH <sub>2</sub> Ph	0
f	Н	0
g	CH <sub>3</sub>	0
h	CH <sub>2</sub> CH=CH <sub>2</sub>	0
i	CO <sub>2</sub> CH <sub>2</sub> Ph	$(OCH_3)_2$

Compound **8a** was debenzylated to **8b** by hydrogenation over Pd/C-catalyst in acetic acid. Finally, the structure elucidation of **8d** and **8a** by X-ray analyses ensured that the oxazolidine derivative had in fact been formed. This result eliminates structure **9** and that of its conceivable precursor **4e**.

#### **Crystal structure discussion**

Eventually, the structure of compounds **8a** and **8d** was proven by X-ray analysis (Figures 1 and 2).

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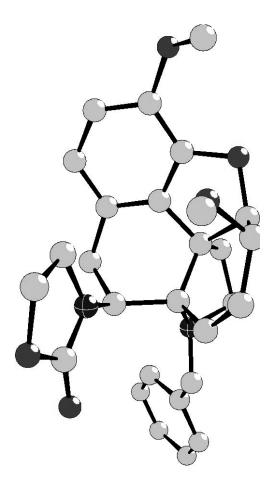


Figure 1. Perspective view of compound 8a.

The crystal data are summarized in the Experimental part. Lists of structure factors, atomic coordinates and thermal components for non-hydrogen atoms and hydrogen atom parameters are available from T.H.B. and have been deposited at the Cambridge Crystallograpgic Data Centre. All bond lengths and angles in the molecules **8a** and **8d** lie within the expected ranges. The phenyl rings (C1,C2,C3,C4,C11,C12) and (C24-C29) are planar, whereas the other six-membered rings are characterized by classical conformations (chair, boat, twist-boat) with deviations of some ring atoms. The five-membered rings (O2,C5,C13,C12,C4) and (C13,C14,N1,C16,C15) have envelope conformations with C5 and C16, respectively, deviating from the planes defined by the other ring atoms. The oxazolidine ring (N2,C17,O4,C18,C19) adopts an envelope conformation only in compound **8a**, where C18 deviates from the ring plane. In structure **8d** this ring is planar.

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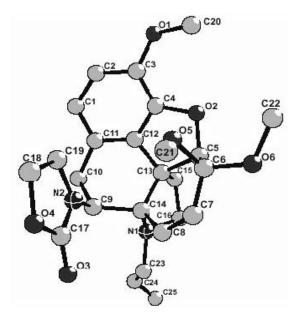


Figure 2 Perspective view of compound 8d with atom labeling scheme.

## **Experimental Section**

General Procedures. Melting points: Kofler hot stage microscope; column chromatography was performed on silica gel (Kieselgel 60, 70–230 mesh; Merck). IR spectra were recorded on a Perkin Elmer Spectrum 1000 instrument; NMR spectra were recorded on a Varian Unity Plus 300 spectrometer; mass spectra were measured on a Shimadzu QP-5000 mass spectrometer (EI, 70eV). Elemental analyses were performed by the Laboratory of Microanalysis, Institute for Physical Chemistry of the University of Vienna. Energy computations were accomplished using the CS Chem 3D Pro® program (MM2 force field), Cambridge, MA.

(7*R*)-*N*-Benzyl-*N*-nor-7-bromoneopinone dimethyl acetal (2). Benzyl bromide (0.75 mL, 6.3 mmol) was added to 14,17-cyclonorcodeinone dimethyl acetal (1, 2.0 g, 6.11 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) and stirred at 25 °C for 6 h. After addition of a saturated aqueous solution of NaHCO<sub>3</sub> (5 mL) stirring was continued for 1 h. The CH<sub>2</sub>Cl<sub>2</sub> layer was separated, washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The residual pale oil **2** (3.0 g, 98%) was purified over silica gel (petrol ether/ethyl acetate/triethylamine 10:2:1, R<sub>f</sub> = 0.43). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.45–7.20 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 6.80 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-2), 6.55 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-1), 5.77 (d,  $J_{7,8}$  = 6.3 Hz, 1H, H-8), 5.22 (s, 1H, H-5), 4.59 (d,  $J_{7,8}$  = 6.3 Hz, 1H, H-7), 3.90 (s, 3H, 3-OCH<sub>3</sub>), 4.01, 3.92 (AB, $J_{gem}$  = 13.4 Hz, 2H, benzyl-HaHb), 3.54 (d,  $J_{9,10\beta}$  = 7.1 Hz, 1H, H-9), 3.53 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.21 (d,  $J_{gem}$  = 18.2 Hz, 1H, H $\alpha$ -10), 3.03 (m, 1H, Ha-16), 3.02 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 2.86 (dd,  $J_{9,10\beta}$  = 7.1 Hz,  $J_{gem}$  = 18.2 Hz, 1H, H $\beta$ -10), 2.68 (m, 1H, Ha-16), 2.50 (m, 1H, Ha-15), 1.69 (m, 1H, HB-15); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 145.68 (C-4), 143.80 (C-14), 142.01 (C-3), 138.91 (C<sub>Ph</sub>-1), 131.25 (C-12), 128.68, 128.20 (C<sub>Ph</sub>-2,6, C<sub>Ph</sub>-3,5), 127.04 (C-11), 126.89 (C<sub>Ph</sub>-4), 119.04 (C-1), 117.00 (C-8), 114.60 (C-2), 98.48 (C-6), 91.89 (C-5), 57.98 (C-9),

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57.02 (benzyl-CH<sub>2</sub>), 57.02 (3-OCH<sub>3</sub>), 50.42, 49.50 ( $6\alpha$ - and  $6\beta$ -OCH<sub>3</sub>), 46.69 (C-13), 46.61 (C-7), 43.46 (C-16), 34.79 (C-15), 31.86 (C-10); MS: m/z 497, 499 (1:1). Anal Calcd for C<sub>26</sub>H<sub>28</sub>BrNO<sub>4</sub> (498.42): C, 62.66; H, 5.66; Br, 16.03; N, 2.81. Found: C, 62.39; H, 5.65; Br, 15.95; N, 2.81.

(9S)-17-Benzyl-4,5-epoxy-9-diethylamino-3-methoxyhasubanan-6-one dimethyl acetal (3b). To a solution of 2 (580 mg, 1.16 mmol) in DMF (15 mL) was added dry diethylamine (0.6 mL, excess), and the mixture was stirred under Ar at 120 °C for 4 h. After evaporation and flash chromatography (petroleum ether/ethyl acetate/triethylamine 12:2:0.5) **3b** (150 mg, 26%) was obtained as an analytically pure oil. Another pure fraction was 3a (120 mg, 22%; formed from dimethylamine present in purchased and unpurified diethylamine; Merck 803010, "z. Synthese"). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.38 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 6.73 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-2), 6.67 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-1), 6.26 (d,  $J_{7,8} = 11.1$  Hz, 1H, H-8), 5.88 (d,  $J_{7,8} = 11.1$  Hz, 1H, H-7), 4.64 (s, 1H, H-5), 4.32 (d,  $J_{gem} = 13.5$  Hz, 1H, benzyl-Ha), 3.98 (s, 3H, 3-OCH<sub>3</sub>), 3.58 (m, 1H, H-9), 3.56 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.49 (d,  $J_{gem} = 13.5$  Hz, 1H, benzyl-HB), 3.11 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 2.99 (d,  $J_{\text{gem}} = 15.6 \text{ Hz}$ , 1H, H $_{\alpha}$ -10), 2.87 (dd,  $J_{9,10\beta} = 4.95 \text{ Hz}$ ,  $J_{\text{gem}} = 15.6 \text{ Hz}$ , 1H, H $_{\beta}$ -10), 2.64 (m, 1H, Ha-16), 2.27 (m, 1H, HB-16), 2.11 (broad m, 4H, 2 x CH<sub>2</sub>CH<sub>3</sub>), 2.02 (m, 1H, Ha-15), 1.62 (m, 1H, H<sub>B</sub>-15), 0.91 (t, 6H, 2 x CH<sub>2</sub>CH<sub>3</sub>);  ${}^{13}$ C NMR (CDCl<sub>3</sub>):  $\delta$  146.21 (C-4), 141.48 (C-3), 140.63 (Cph-1), 133.74 (C-8), 132.19 (C-12), 128.26 (Cph-2,6), 128.08 (Cph-3,5), 127.44 (C-11), 126.82 (Cph-4), 125.29 (C-7), 118.79 (C-1), 115.23 (C-2), 94.49 (C-6), 94.36 (C-5), 67.06 (C-9), 65.97 (C-14), 57.47 (3-OCH<sub>3</sub>), 53.23 (benzyl-CH<sub>2</sub>), 52.39 (C-13), 50.08, 48.78 ( $6\alpha$ -,  $6\beta$ -OCH<sub>3</sub>), 46.55 (C-16), 44.29 (CH<sub>2</sub>CH<sub>3</sub>), 40.75 (C-15), 23.64 (C-10), 15.00 (CH<sub>2</sub>CH<sub>3</sub>); MS: m/z 490,  $C_{30}H_{38}N_2O_4$  (490.63).

(9S)-17-benzyl-4,5-epoxy-3-methoxy-9-dimethylaminohasubanan-6-one dimethyl acetal (3a). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.37 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 6.72 (d, *J*<sub>12</sub>= 8.0 Hz, 1H, H-2), 6.69 (d, *J*<sub>12</sub>= 8.0 Hz, 1H, H-1), 6.21 (d, *J*<sub>7,8</sub>= 11.1 Hz, 1H, H-7), 5.93 (d, *J*<sub>7,8</sub>= 11.1 Hz, 1H, H-8), 4.65 (s, 1H, H-5), 4.28 (d, *J*<sub>gem</sub> = 13.5 Hz, 1H, benzyl-Ha), 3.97 (s, 3H, 3-OCH<sub>3</sub>), 3.56 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 3.50 (d, *J*<sub>gem</sub> = 13.5 Hz, 1H, benzyl-HB), 3.24 (m, 1H, H-9), 3.09 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 2.97 (m, *J*<sub>gem</sub> = 15.6 Hz, 2H, H-10), 2.68 (m, 1H, Hα-16 or Hβ-16), 2.31 (m, 1H, Hα-16 or Hβ-16), 2.05 (m, 1H, Hα-15 or Hβ-15), 1.97 (s, 6H, 2 x CH<sub>3</sub>), 1.62 (m, 1H, H-15α or H-15β); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 146.20 (C-4), 141.46 (C-3), 140.47 (C<sub>Ph</sub>-1), 133.31 (C-8), 132.19 (C-12), 128.29 (C<sub>Ph</sub>-2,6), 127.98 (C<sub>Ph</sub>-3,5), 127.14 (C-11), 126.85 (C<sub>Ph</sub>-4), 126.76 (C-7), 119.05 (C-1), 115.25 (C-2), 94.72 (C-6), 94.62 (C-5), 69.90 (C-9), 65.88 (C-14), 57.44 (3-OCH<sub>3</sub>), 52.82 (benzyl-CH<sub>2</sub>), 52.16 (C-13), 50.19, 48.27 (6α-, 6β-OCH<sub>3</sub>), 46.69 (C-16), 43.12 (CH<sub>3</sub>), 41.49 (C-15), 21.61 (C-10); MS: *m/z* 462, C<sub>30</sub>H<sub>34</sub>N<sub>2</sub>O<sub>4</sub> (462.59).

(9S)-17-Benzyl-4,5-epoxy-9-hydroxyethylamino-3-methoxyhasubanan-6-one dimethyl acetal (3c). A solution of 2 (6.08 g, 12.21 mmol) and aminoethanol (7.4 mL, excess) in dry DMF (100 mL) was stirred under Ar at 130 °C (bath temperature) for 4 h. After evaporation of the solvent the residue was taken up with ethyl acetate. The organic extract was washed with brine and water, dried with Na<sub>2</sub>SO<sub>4</sub>, and after removal of the solvent under reduced pressure 3c was obtained a pale orange oil (5.73 g, 98%), which was sufficiently pure for further reactions. An analytical sample was crystallized from ethanol and gave colourless crystals, mp

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158–160 °C; ¹H NMR (CDCl<sub>3</sub>): δ 7.37–7.27 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 6.74 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-1), 6.73 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-2), 6.19 (d,  $J_{7,8}$  = 10.8 Hz, 1H, H-7), 6.00 (d,  $J_{7,8}$  = 10.8 Hz, 1H, H-8), 4.62 (s, 1H, H-5), 4.27 (d,  $J_{gem}$  = 13.5 Hz, 1H, benzyl-Ha), 3.95 (s, 3H, 3-OCH<sub>3</sub>), 3.59 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 3.52 (d, 1H, benzyl-HB), 3.45, 3.25 (m, m, 1H, 1H, OCHAHB), 3.21 (d,  $J_{gem}$  = 14.7 Hz, 1H, Hα-10), 3.19 (broad s, 1H, H-9), 3.04 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 2.88 (broad d, 1H, Hβ-10), 2.75 (m, 1H, Hax-16), 2.66, 2.48 (m, m, 1H, 1H, NCHAHB), 2.34 (m, 1H, Heq-16), 2.12 (m, 1H, Hax-15), 1.60 (m, 1H, Heq-15). <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 146.12 (C-4), 141.78 (C-3), 140.03 (CPh-1), 133.16 (C-12), 132.38, 130.58 (C-7, C-8), 128.23, 127.28 (CPh-2,6, CPh-3,5), 126.83 (CPh-4), 123.64 (C-11), 120.53 (C-1), 114.42 (C-2), 95.46 (C-6), 94.63 (C-5), 68.44 (C-14), 60.16 (OCH<sub>2</sub>), 59.63 (C-9), 57.17 (3-OCH<sub>3</sub>), 53.09 (benzyl-CH<sub>2</sub>), 52.27 (C-13), 50.82, 48.49 (6α- and 6β-OCH<sub>3</sub>), 47.91 (C-16), 47.43 (NCH<sub>2</sub>), 41.86 (C-15), 27.45 (C-10); MS: m/z (%) 478 (0.35), 446 (100; M –CH<sub>3</sub>OH). Anal. Calcd. for C<sub>28</sub>H<sub>34</sub>N<sub>2</sub>O<sub>5</sub> (478.59): C, 70.27; H, 7.16; N, 5.85. Found: C, 70.26; H, 7.30; N, 5.78.

## (9S)-17-Benzyl-(N-benzyloxycarbonyl)-4,5-epoxy-9-hydroxyethylamino-3-meth-

oxyhasubanan-6-one dimethyl acetal (3d). To a solution of 3c (3.10 g, 6.5 mmol) in acetone (65 mL) were added K<sub>2</sub>CO<sub>3</sub> (1.0 g, 7.2 mmol) and benzyl chloroformate (1mL, 1.2g, 7.0 mmol), and the mixture was stirred at 40 °C (bath temperature) under Ar for 2 h. After concentration, the crude residue (3.81 g, 96%) was found to be sufficiently pure for further reactions. For analytical purpose a sample was purified by flash chromatography on silica gel (petrol ether/ethyl acetate/triethylamine 5:2:1). The NMR spectra show two sets of spectra due to the presence of amide rotamers. <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.33–7.08 (m, 10H, 2 C<sub>6</sub>H<sub>5</sub>), 6.76–6.63 (AB, 2H, H-1, H-2), 5.93–5.71 (m, 2H, H-7, H-8), 5.26–5.04 (m, 2H, benzyl-CH<sub>2</sub>), 4.99, 4.82 (d, d, 1H, 1H, H-9), 4.60, 4.56 (s, s, 1H, 1H, H-5), 4.19 (d, 1H,  $J_{gem} = 13.7$  Hz, benzyl-Ha), 3.92 (s, 3H, 3-OCH<sub>3</sub>), 3.50 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.46 – 3.12 (m, 5H, HA-10, CH<sub>2</sub>CH<sub>2</sub>O), 3.05 (d, 1H,  $J_{gem} = 13.7$ Hz, benzyl-HB), 3.02, 2.96 (s, s, 3H, 3H, 2  $6\alpha$ - and  $6\beta$ -OCH<sub>3</sub>), 2.85–2.52 (m, 1H, HB-10, HA-16), 2.30–1.79 (m, 1H, H-15A, HB-16), 1.65–1.45 (m, 1H, H-15B).  $^{13}$ C NMR (CDCl<sub>3</sub>):  $\delta$  158.49, 158.00 (2 CO), 146.49, 126.41 (2 C-4), 142.26 (C-3), 140.39, 139.90 (2 N-benzyl-Cph-1), 136.46, 135.11 (2 O-benzyl-C<sub>Ph</sub>-1), 132.92, 132.12 (2 C-12), 130.75, 130.48, 130.14 (C-7, C-8), 128.63, 128.48, 128.41, 128.17, 128.09, 127.90, 127.77 (Cph), 126.67 (2 Cph-4), 124.45, 123.86 (2 C-11), 119.02, 118.49 (2 C-1), 114.87, 114.65 (2 C-2), 94.43, 94.15 (2 C-5), 94.27 (C-6), 67.85 (C-14), 68.57, 67.26 (2 benzyl-CH<sub>2</sub>O), 64.23, 62.03 (2 CH<sub>2</sub>O), 60.53, 60.12 (2 C-9), 57.24  $(3-OCH_3)$ , 52.40, 52.03 (CH<sub>2</sub>N, benzyl-CH<sub>2</sub>N), 50.31, 48.55 ( $6\alpha$ -,  $6\beta$ -OCH<sub>3</sub>), 46.76, 46.42 (2 C-16), 46.42, 45.15 (2 C-13), 41.21, 40.95 (2 C-15), 29.21, 28.99 (2 C-10); MS: m/z 612. Anal. Calcd. for C<sub>36</sub>H<sub>40</sub>N<sub>2</sub>O<sub>7</sub> (612.72): C, 70.57; H, 6.58; N, 4.57. Found: C, 70.31; H, 6.62; N, 4.65. (9S)-17-Benzyl-4,5-epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one

acetal (8a). A mixture of 3d (10.3 g, 16.8 mmol), CH<sub>2</sub>Cl<sub>2</sub> (120 mL), *p*-toluenesulfonyl chloride (3.2 g, 17 mmol, freshly purified), 4-dimethylaminopyridine (2.1 g, 17.2 mmol) and triethylamine (5 mL, 3.6 g, 36 mmol) was stirred under Ar at 40 °C (bath temperature) for 10 h. After evaporation of the solvent the residue was dissolved in ethyl acetate, the organic extract was washed with a saturated aqueous NaHCO<sub>3</sub> solution, then with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and

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concentrated under reduced pressure. After purification of the residue by flash chromatography on silica gel (petrol ether/ethyl acetate/triethylamine 6:4:1) colourless crystals **8a** were isolated (7.95 g, 94%); mp 188–190 °C (MeOH); ¹H NMR (CDCl<sub>3</sub>): δ 7.39–7.26 (m, 5H,  $C_6H_5$ ), 6.75 (d,  $J_{1,2} = 8.2$  Hz, 1H, H-1), 6.71 (d,  $J_{1,2} = 8.2$  Hz, H, H-1), 6.20 (d,  $J_{7,8} = 10.8$  Hz, 1H, H-7), 6.02 (d,  $J_{7,8} = 10.8$  Hz, 1H, H-8), 4.67 (d,  $J_{gem} = 12$  Hz, 1H, benzyl-Ha), 4.66 (s, 1H, H-5), 4.57 (m, 1H, H-9), 4.55–3.92 (m, 2H, HaHb-5¹), 3.97 (s, 3H, 3-OCH<sub>3</sub>), 3.55 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.46 (d,  $J_{gem} = 12$  Hz, 1H, benzyl-Hb), 3.49, 2.82 (m, m, 1H, 1H, HaHb-10), 3.09 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 2.93, 2.65 (m, m, 1H, 1H, HaHb-4¹), 2.77, 2.30 (m, m, 1H, 1H, HaHb-16), 2.09, 1.63 (m, m, 1H, 1H, HaHb-15). ¹³C NMR (CDCl<sub>3</sub>): δ 159.11 (C-2¹), 146.08 (C-4), 142.09 (C-3), 140.07 (CPh-1), 132.44 (C-12), 130.84, 130.60 (C-7, C-8), 128.11 (CPh-2,3,5,6), 126.77 (CPh-4), 124.39 (C-11), 119.51 (C-1), 115.09 (C-2), 94.47 (C-6), 94.23 (C-5), 67.95 (C-14), 62.18 (C-5¹), 57.23 (3-OCH<sub>3</sub>), 57.13 (C-9), 52.70 (benzyl-CH<sub>2</sub>), 51.75 (C-13), 50.36, 48.61 ( $6\alpha$ -,  $6\beta$ -OCH<sub>3</sub>), 47.05 (C-16), 42.14 (C-4¹), 41.78 (C-15), 27.90 (C-10); IR (KBr): 1747.8, 1722.3 cm-1; MS: m/z 504. Anal Calcd for  $C_{29}H_{32}N_2O_6$  (504.54): C, 69.03; H, 6.39; N, 5.55. Found C, 68.83; H, 6.49; N, 5.38.

## X-ray analysis of compound 8a

With methylene chloride as solvent inclusion ( $C_{29}H_{32}N_2O_6 + CH_2Cl_2$ , M=504.54+84.93). The compound crystallized in monoclinic space group  $P2_1$  with unit cell parameters a = 9.440(2), b = 10.059(2), c = 14.889(3) Å,  $\beta = 95.51(3)^\circ$ , V = 1407.3(5) Å3, Z = 2,  $D_c = 1.391$  g/cm-3,  $\mu = 0.278$  mm-1. Mo-K $\alpha$  radiation ( $\lambda = 0.71073$  Å), 5748 reflections measured, 5738 unique, 5549 with  $F = 2\sigma(F)$  gave  $R1 = 0.029/wR_2$  (all data) = 0.073 in a full matrix least squares refinement with 498 parameters. Goodness-of-fit on  $F_2 = 1.073$ , absolute structure parameter FLACK = 0.02(4). The crystal structure data of **8a** have been deposited at the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; deposition number 155910.

(*9S*)-4,5-Epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one dimethyl acetal (*8b*). A solution of **8a** (1.45 g, 2.9 mmol) and Pd/C-hydrogenation catalyst (1.5 g, 5% Pd; Sigma) in acetic acid (50 mL) was vigorously stirred under hydrogen gas at 1 bar at room temperature for 1h. After filtration through celite on a glass suction sinter funnel the filtrate was concentrated to afford a pale oil of fair purity (1.08 g, 91%), which crystallized from ethyl acetate; mp 208–211°C; IR (KBr): 1751.8 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 6.70 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-1), 6.66 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-2), 5.87 (d,  $J_{7,8}$  = 10.5 Hz, 1H, H-8), 5.79 (d,  $J_{7,8}$  = 10.5 Hz, 1H, H-7), 4.60 (s, 1H, H-5), 4.34 (dd,  $J_{9,10}$  = 3.0 Hz,  $J_{9,10}$  = 2.3 Hz, 1H, H-9), 4.06–3.85 (m, 2H, HaHb-5'), 3.90 (s, 3H, 3-OCH<sub>3</sub>), 3.48 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 3.33 (dd,  $J_{gem}$  = 15.9 Hz,  $J_{9,10}$  = 3.0 Hz, 1H, Ha-10), 3.03 (s, 3H, 6α- or 6β-OCH<sub>3</sub>), 2.97–2.91 (m, 1H, Ha-16), 2.88–2.80 (m, 1H, Ha-4'), 2.81–2.75 (m, 1H, HB-16), 2.73–2.66 (m, 1H, HB-10), 2.58–2.51 (m, 1H, HB-4'), 2.17-2.11 (dd,  $J_{gem}$  = 12.3 Hz, 1H, HA-15), 1.72–1.58 (m, 1H, HB-15). <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 158.95 (C-2'), 146.11 (C-4), 141.95 (C-3), 135.03 (C-8), 131.95 (C-12), 127.93 (C-7), 124.25 (C-11), 119.55 (C-1), 114.95 (C-2), 94.80 (C-6), 94.11 (C-5), 66.12 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-14), 62.03 (C-5'), 59.72 (C-9), 57.07, 50.20, 48.50 (3-12), 143.95 (C-1

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OCH<sub>3</sub>,  $6\alpha$ - and  $6\beta$ -OCH<sub>3</sub>), 50.10 (C-13), 43.98 (C-15), 41.91, 41.87 (C-16, C-4'), 27.19 (C-10); MS: m/z 414.46. Anal. Calcd. for  $C_{22}H_{26}N_2O_6$  (414.46): C, 63.76; H, 6.32; N, 6.76. Found: C, 63.64; H, 6.39; N, 6.70.

(9S)-4,5-Epoxy-3-methoxy-17-methyl-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one dimethyl acetal (8c). MeI (1 mL, 2.28 g, 16 mmol) was added to a solution of 8b (410 mg, 1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL), and the mixture was stirred under Ar at room temperature for 24 h. After concentrating the solution, the residue was dissolved in ethyl acetate. The solution was placed in a separatory funnel and washed with 2 M NaOH and subsequently with brine. The organic layer was separated and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent, the residue was subjected to flash chromatography on silica gel (petrol ether/ethyl acetate/triethylamine 5:2:1,  $R_f = 0.35$ ) affording 8c (240 mg, 57%), which crystallized from methanol; mp 195–197 °C; IR (KBr): 1745.6 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  6.68 (d,  $J_{1,2}$  = 8.2 Hz, 1H, H-2), 6.63 (d,  $J_{1,2}$  = 8.2 Hz, 1H, H-1), 5.97 (d,  $J_{7,8} = 11.1$  Hz, 1H, H-8), 5.91 (d,  $J_{7,8} = 11.1$  Hz, 1H, H-7), 4.57 (s, 1H, H-5), 4.35 (dd, 1H, H-9), 4.05-3.82 (m, 2H, HaHb-5'), 3.91 (s, 3H,  $3-OCH_3$ ), 3.49 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.25 (dd,  $J_{gem} = 15.6 \text{ Hz}$ ,  $J_{9,10} = 4.5 \text{ Hz}$ , 1H, Ha-10), 3.01 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 2.87–2.77 (m, 2H, Ha-4', Ha-16), 2.66 (dd,  $J_{9,10} = 2.4$  Hz, 1H, Hb-10), 2.58 (s, 3H, NCH<sub>3</sub>), 2.57–2.50 (m, 1H, H<sub>B</sub>-4'), 2.42–2.33 (m, 1H, H<sub>B</sub>-16), 2.09–2.04 (m, 1H, H<sub>A</sub>-15), 1.67–1.57 (m, 1H, H<sub>B</sub>-15). <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 158.99 (C-2'), 146.08 (C-4), 142.03 (C-3), 132.35 (C-12), 130.28 (C-7, C-8), 124.53 (C-11), 119.39 (C-1), 115.08 (C-2), 94.31 (C-6), 94.15 (C-5), 67.39 (C-14), 62.11 (C-5'), 57.22 (3-OCH<sub>3</sub>), 56.76 (C-9), 51.59 (C-13), 50.82 (C-16), 50.28 (6α- or 6β-OCH<sub>3</sub>), 48.51 (6α- or 6β-OCH<sub>3</sub>), 42.03 (C-4'), 41.47 (C-15), 35.27 (NCH<sub>3</sub>), 27.82 (C-10). MS: m/z 428. Anal. Calcd. for C<sub>23</sub>H<sub>28</sub>N<sub>2</sub>O<sub>6</sub> (428.48): C, 64.47; H, 6.59; N, 6.54. Found: C, 64.34; H, 6.63; N, 6.41.

17-Allyl-4,5-epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one acetal (8d). To a solution of 8b (320 mg, 0.8 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added 3bromopropene (0.75 mL, 1.05g, 8.7 mmol), and the mixture was stirred under Ar at room temperature for 48 h. After concentrating, the mixture was taken up in ethyl acetate, placed in a separatory funnel, and sodium carbonate (2 mL, 2 M aqueous solution) was added. Washing the organic layer with brine, drying over Na<sub>2</sub>SO<sub>4</sub> and evaporation of the solvent furnished a crude oil (240 mg), which was purified by flash chromatography on silica gel (petrol ether/ethyl acetate/triethylamine 3:2:1, Rf 0.54) to yield crystalline 8d (160 mg, 46%); mp 158–160 °C (ethyl acetate); IR (KBr):1736.3cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  6.00 (d,  $J_{1,2} = 8.1$  Hz, 1H, H-2), 5.89  $(d, J_{1,2} = 8.1 \text{ Hz}, 1H, H-1), 6.01 (d, J_{7,8} = 11.1 \text{ Hz}, 1H, H-8), 5.89 (d, J_{7,8} = 11.1 \text{ Hz}, 1H, H-7),$ 5.90-5.77 (m, 1H, CH=), 5.18 (d,  $J_{vic} = 16.8$  Hz, 1H, =CHA), 5.06 (d,  $J_{vic} = 9.9$  Hz, 1H, =CHB), 4.57 (s, 1H, H-5), 4.33 (dd, 1H, H-9), 4.05–3.82 (m, 3H, HaHb-5', NCHa), 3.90 (s, 3H, 3-OCH<sub>3</sub>), 3.48 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 3.15 (dd, 1H, HA-10), 3.01 (s, 3H,  $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 2.99–2.70 (m, 3H, Ha-4', Ha-16, NCHB), 2.65 (dd, 1H, HB-10), 2.60–2.45 (m, 1H, HB-4'), 2.25–2.18 (m, 1H, H<sub>B</sub>-16), 2.08–2.01 (m, 1H, H<sub>A</sub>-15), 1.64–1.53 (m,1H, H<sub>B</sub>-15); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 159.02 (C-2'), 146.01 (C-4), 141.99 (C-3), 137.05 (=CH), 132.38 (C-12), 130.64, 130.18 (C-7, C-8), 124.44 (C-11), 119.38 (C-1), 115.93 (CH2), 115.05 (C-2), 94.39 (C-6), 94.12 (C-5), 67.71 (C-14), 62.11 (C-5'), 57.18 (3-OCH<sub>3</sub>), 57.03 (C-9), 51.70 (C-13), 51.41 (NCH<sub>2</sub>), 50.27 ( $6\alpha$ - or  $6\beta$ -OCH<sub>3</sub>), 48.51

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(6α- or 6β-OCH<sub>3</sub>), 47.15 (C-16), 42.04 (C-4', 41.57 (C-15), 27.74 (C-10). MS: m/z 453. Anal. Calcd. for  $C_{25}H_{30}N_2O_6$  (454.52): C, 66.06; H, 6.65; N, 6.16. Found: C, 65.98; H, 6.77; N, 6.21. **X-ray analysis of 8d** ( $C_{25}H_{30}N_2O_6$ , M = 454.52): The compound crystallized in orthorhombic space group  $P2_12_12_1$  with unit cell parameters a = 9.4720(1), b = 11.8370(2), c = 19.8010(3) Å, α =  $\beta = \gamma = 90^\circ$ , V = 2220.09(6) Å<sub>3</sub>, Z = 4, D<sub>c</sub> = 1.360 g cm-3,  $\mu = 0.090$  mm-1. Mo-Kα radiation ( $\lambda = 0.71073$  Å), 6736 reflections measured, 6711 unique, 6098 with F = 2σ(F) gave R1 = 0.036 / wR<sub>2</sub> (all data) = 0.0827 in a full matrix least squares refinement with 419 parameters. Goodness-of-fit on F<sub>2</sub> = 1.027, absolute structure parameter FLACK = 0.0(6). The crystal structure of **8d** has been deposited at the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; deposition number 155909.

(9S)-17-Benzyl-4,5-epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one (8e). Α solution of 8a (1.22 g, 2.4 mmol) in EtOH (10 mL) and 1 N HCl (3 mL) was heated for 5 min. After cooling the mixture was made alkaline with 2 M NaOH (4 mL) and extracted with ethyl acetate. The extract was washed with water and dried over Na<sub>2</sub>SO<sub>4</sub> to yield colourless crystals 8e (1.07 g, 97%); mp 134–136 °C (MeOH); IR (KBr): 1724.1, 1711.2 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.36-7.27 (m, 5H,  $C_6H_5$ ), 7.04 (d,  $J_{7,8} = 10.8$  Hz, 1H, H-8), 6.77 (d,  $J_{1,2} = 8.1$  Hz, 1H, H-1), 6.74 $(d, J_{1,2} = 8.1 \text{ Hz}, 1H, H-2), 6.28 (d, J_{1,2} = 8.1 \text{ Hz}, 1H, H-7), 4.84 (s, 1H, H-5), 4.68 (d, J_{gem} = 13.2)$ Hz, 1H, benzyl-Ha), 4.68 (m, 1H, H-9), 3.97–3.77 (m, 2H, HaHb-5'), 3.92 (s, 3H, 3-OCH<sub>3</sub>), 3.53  $(d, J_{gem} = 13.2 \text{ Hz}, 1H, \text{benzyl-HB}), 3.50 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HB-10, HA-16), 2.63 (m, 1H, HA-10), 2.91 (m, 2H, HA-10), 2.9$ 1H, Ha-4'), 2.51 (m, 1H, Hb-16), 2.22 (m, 1H, Ha-15), 1.96 (m, 1H, Hb-4'), 1.77 (m, 1H, Hb-15). <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 195.16 (C-6), 159.02 (C-2'), 148.37 (C-8), 144.80 (C-4), 143.38 (C-3), 139.20 (Cph-1), 130.02 (C-12), 129.59 (C-7), 128,40, 128.10 (Cph-2,3,5,6), 127.19 (Cph-4), 124.12 (C-11), 120.85 (C-1), 114.36 (C-2), 89.71 (C-5), 67.82 (C-14), 62.03 (C-5'), 57.11 (C-9), 56.58 (3-OCH<sub>3</sub>), 54.07 (C-13), 52.85 (benzyl-CH<sub>2</sub>), 48.13 (C-16), 41.76 (C-4'), 28.47 (C-10); MS: m/z 458. Anal. Calcd. for  $C_{27}H_{26}N_2O_5$  (458.51): C, 70.73; H, 5.72; N, 6.11. Found: C, 70.51; H, 5.81; N, 5.98.

(9*S*)-4,5-Epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one (8*f*). A solution of 8b (550 mg, 1.3 mmol) in 1 N HCl (4 mL) was heated for 5 min. After cooling, the mixture was made alkaline with 2 M NaOH (3 mL) and extracted with ethyl acetate. The organic extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and after evaporation of the solvent pale yellow crystals 8f (470 mg, 96%) were obtained; mp 213–215 °C (MeOH); IR (KBr): 3440.3, 3296.9, 1745.1, 1704.4 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 6.73 (s, 2H, H-1, H-2), 6.68 (d,  $J_{7,8}$  = 10.8 Hz, 1H, H-8), 6.09 (d,  $J_{7,8}$  = 10.8 Hz, 1H, H-7), 4.81 (s, 1H, H-5), 4.46 (m, 1H, H-9), 3.90 (s, 3H, 3-OCH<sub>3</sub>), 3.93–3.70 (m, 2H, HaH<sub>B</sub>-5'), 3.34 (dd,  $J_{gem}$  = 15.9 Hz,  $J_{9,10}$  = 4.2 Hz, 1H, Ha-10), 3.14–2.95 (m, 2H, HaH<sub>B</sub>-16), 2.72 (dd,  $J_{9,10}$  = 2.4 Hz, 1H, H<sub>B</sub>-10), 2.58–2.42 (m, 1H, Ha-4'), 2.29 (dd,  $J_{gem}$  = 12.6 Hz,  $J_{15A,16}$  = 3.9 Hz, 1H, Ha-15), 1.96-1.87 (m, 1H, H<sub>B</sub>-4'), 1.86–1.77 (m, 1H, H<sub>B</sub>-15). <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 195.23 (C-6), 158.88 (C-2'), 150.32 (C-8), 144.99 (C-4), 143.29 (C-3), 129.67 (C-12), 126.73 (C-7), 124.10 (C-11), 120.88 (C-1), 114.24 (C-2), 89.40 (C-5), 65.73 (C-14), 61.95 (C-5'), 59.67 (C-9), 56.50 (3-OCH<sub>3</sub>), 52.42 (C-13), 43.17 (C-15, C-16), 41.54 (C-4'), 27.95 (C-10); MS: m/z 368. Anal. Calcd. for C<sub>20</sub>H<sub>20</sub>N<sub>2</sub>O<sub>5</sub> (368.39): C, 65.21; H, 5.47; N, 7.60. Found: C,

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65.01; H, 5.64; N, 7.41.

- (*9S*)-4,5-Epoxy-3-methoxy-17-methyl-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one (*8g*). A solution of *8c* (100 mg, 0.23 mmol) in 1 N HCl (3 mL) was heated for 5 min. After cooling the mixture was made alkaline with 2 M NaOH (3 mL) and extracted with ethyl acetate. The extract was washed with brine and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent colourless crystals *8g* (80 mg, 90%) were obtained; mp 209–211 °C (ethyl acetate); IR (KBr): 1724.2, 1706.6 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 6.93 (d, *J*<sub>7,8</sub> = 10.8 Hz, 1H, H-8), 6.71 (s, 2H, H-1, H-2), 6.20 (d, *J*<sub>7,8</sub> = 10.8 Hz, 1H, H-7), 4.78 (s, 1H, H-5), 4.48 (dd, 1H, H-9), 3.89 (s, 3H, 3-OCH<sub>3</sub>), 3.92–3.70 (m, 2H, HaHb-5'), 3.26 (dd, *J*<sub>gem</sub> = 15.6 Hz, *J*<sub>9,10</sub> = 4.5 Hz, 1H, Ha-10), 2.96 (dd, 1H, Ha-16), 2.70 (dd, *J*<sub>gem</sub> = 15.6 Hz, *J*<sub>9,10</sub> = 2.4 Hz, 1H, HB-10), 2.67 (s, 3H, NCH<sub>3</sub>), 2.64 (m, 1H, HB-16), 2.57 (m, 1H, Ha-4'), 2.21 (m, 1H, Ha-15), 1.91 (m, 1H, HB-4'), 1.79 (m, 1H, HB-15); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 195.04 (C-6), 158.86 (C-2'), 148.28 (C-8), 144.71 (C-4), 143.21 (C-3), 129.98 (C-12), 129.36 (C-7), 124.22 (C-11), 120.70 (C-1), 114.21 (C-2), 89.59 (C-5), 67.27 (C-14), 61.89 (C-5'), 56.75 (C-9), 56.49 (3-OCH<sub>3</sub>), 53.82 (C-13), 51.87 (C-16), 41.62 (C-4'), 40.17 (C-15), 35.42 (17-NCH<sub>3</sub>), 28.32 (C-10); MS: *m*/*z* 382. Anal. Calcd. for C<sub>21</sub>H<sub>22</sub>N<sub>2</sub>O<sub>5</sub> (382.42): C, 65.96; H, 5.80; N, 7.33. Found: C, 65.69; H, 5.97; N, 7.17.
- (9S)-17-Allyl-4,5-epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasubanan-6-one (8h). A solution of 8d (100 mg, 0.22 mmol) in 1 N HCl (3 mL) was heated for 5 min. After cooling, the mixture was made alkaline with 2 M NaOH (3 mL) and extracted with ethyl acetate. The extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent, 8h (80 mg, 89%) was obtained as solid material; IR (KBr): 1738.4, 1701.2 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  6.90 (d,  $J_{7,8}$  = 10.8 Hz, 1H, H-8), 6.71 (s, 2H, H-1, H-2), 6.18 (d,  $J_{7,8} = 10.8$  Hz, 1H, H-7), 6.22–5.76 (m, 1H, =CH), 5.24 (d,  $J_{\text{vic}} = 17.1$  Hz, 1H, =CHA), 5.12 (d,  $J_{\text{vic}} = 10.2$  Hz; 1H, =CHB), 4.78 (s, 1H, H-5), 4.48 (m, 1H, H-9), 3.98 (m, 1H, NCHa), 3.95–3.68 (m, 2H, HaHb-5'), 3.89 (s, 3H, 3-OCH<sub>3</sub>), 3.31 (dd,  $J_{gem} = 15.7 \text{ Hz}$ ,  $J_{9,10} = 4.5 \text{ Hz}$ , 1H, Ha-10), 3.11–3.04 (m, 2H, Ha-16, NCH<sub>B</sub>), 2.70 (dd,  $J_{\text{gem}} = 15.7 \text{ Hz}$ ,  $J_{9,10} = 2.5 \text{ Hz}$ ,  $J_{1}$ ,  $J_{1}$ ,  $J_{1}$ ,  $J_{2}$ ,  $J_{2}$ ,  $J_{3}$ ,  $J_{2}$ ,  $J_{3}$ ,  $J_$ 15), 1.90 (m, 1H, H<sub>B</sub>-4'), 1.76 (m, 1H, H<sub>B</sub>-15);  ${}^{13}$ C NMR (CDCl<sub>3</sub>):  $\delta$  195.09 (C-6), 158.90 (C-2'), 148.27 (C-8), 144.68 (C-4), 143.21 (C-3), 136.25 (=CH), 129.93 (C-12), 129.14 (C-7), 124.15 (C-11), 120.71 (C-1), 116.79 (=CH<sub>2</sub>), 114.23 (C-2), 89.54 (5), 67.50 (C-14), 61.93 (C-5'), 56.99 (C-9), 56.49 (3-OCH<sub>3</sub>), 54.00 (C-13), 51.49 (NCH<sub>2</sub>), 48.22 (C-16), 41.63 (C-4'), 40.23 (C-15), 28.27 (C-10); MS: m/z 408. Anal. Calcd. for C<sub>23</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub>(408.45): C, 67.63; H, 5.92; N, 6.86. Found: C, 67.54; H, 6.21; N, 6.92.
- (9S) 17-Benzyloxycarbonyl-4,5-epoxy-3-methoxy-9-(oxazolidin-2'-on-3'-yl)hasuban-an-6-one (8i). To a solution of 8b (250 mg, 0.6 mmol) in acetone (20 mL) was added  $K_2CO_3$  (100 mg) and benzyl chloroformate (0.1 mL, 120mg, 0.7 mmol). The mixture was stirred under Ar at room temperature for 10 h. After concentration under reduced pressure the residue was extracted with ethyl acetate. The extract was washed with brine, dried over  $Na_2SO_4$ , and after removal of the solvent an orange oil 8i (330 mg, 100%) was isolated which was purified by flash chromatography on silica gel (petrol ether/ethyl acetate/triethylamine 3:2:1,  $R_f$  0.22); IR (KBr): 1754.4, 1698.5 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.44-7.23 (m, 5H,  $C_6H_5$ ), 6.74 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-

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2), 6.66 (d,  $J_{1,2}$  = 8.1 Hz, 1H, H-1), 6.43 (d,  $J_{7,8}$  = 10.4 Hz, 1H, H-8), 5.79 (d, 1H, H-7), 5.31 (m, 1H, H-9), 5.21 (broad s, 2H, benzyl-HaHB), 4.71 (s, 1H, H-5), 4.10–3.91 (m, 2H, HaHB-5'), 3.92 (s, 3H, 3-OCH<sub>3</sub>), 3.86–3.39 (m, 2H, HaHB-16), 3.29, 3.23 (2 s, 3H, 3H, 6α- and 6β-OCH<sub>3</sub>), 3.48, 2.97 (2 m, 1H, 1H, HaHB-10), 2.95 (m, 2H, HaHB-4'), 2.26-2.03 (m, 2H, HaHB-15); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 158.23 (C-2'), 154.29 (NCO), 146.94 (C-4), 142.65 (C-3), 136.89 (CPh-1), 129.93 (C-12), 129.86, 129.62 (C-7, C-8), 128.32 (CPh-2,3,5,6), 127.85 (CPh-4), 124.66 (C-11), 119.94 (C-1), 116.48 (C-2), 94.82 (C-6), 93.03 (C-5), 66.97 (benzyl-CH<sub>2</sub>), 65.97 (C-14), 62.06 (C-5'), 57.35 (3-OCH<sub>3</sub>), 55.93 (C-9), 54.65 (C-13), 49.39, 48.89 (6α- and 6β-OCH<sub>3</sub>), 44.91 (C-16), 42.29 (C-4'), 38.72 (C-15), 28.84 (C-10); MS: m/z 548,  $C_{30}H_{32}N_{2}O_{8}$  (548.59).

## X-ray structure analysis

**Crystallography. Sample preparation.** Crystals suitable for X-ray investigation were prepared by slow evaporation of solutions of the respective compounds. Colourless crystals were grown from ethyl acetate in the case of **8d.** Crystallization from CH<sub>2</sub>Cl<sub>2</sub> gave the clathrate (inclusion compound) **8a**•CH<sub>2</sub>Cl<sub>2</sub> (1:1).

**Data collection, structure analysis, and refinement.** Intensity data for **8a** and **8d** where obtained on a Nonius Kappa instrument with CCD detector at 110 K (in detail: **8a/8d**, collection time per frame 65/50 sec., rotation width 1.5/1.0 deg., detector distance 35/35 mm, total number of frames 570/392). The structures **8a** and **8d** were solved by direct methods, using the program *SIR92*6 and refined with *SHELXL93*. <sup>7</sup> The figures were prepared using *DIAMOND 2.1a*. <sup>8</sup>

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